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9 Abstract

10 The aim of this study is to compare the quality characteristics of dry-cured loins with 11 different levels of proteolysis and lipid oxidation and to investigate the relationship between 12 these factors on quality characteristics. The dry-cured loins were divided into four groups (HH, 13 HL, LH, and LL) based on the proteolysis index and 2-thiobarbituric acid reactive substances 14 (TBARS). Moisture, protein, and fat content were all significantly influenced by proteolysis and 15 lipid oxidation (p<0.05). The total fatty acid content in the high proteolysis groups (HH and HL) 16 was significantly lower than that in the low proteolysis groups (LH and LL) (p<0.05). For total 17 free amino acid content, HH was the highest, and LL was the lowest (p<0.05). On the other hand, there was no significant difference between HL and LH (p>0.05). In the amount of total volatile 18 19 compounds, there was no significant difference between HH and HL (p>0.05), but LH and LL 20 significantly differed (p<0.05). In conclusion, proteolysis and lipid oxidation can influence the 21 quality characteristics of dry-cured loin. Additionally, proteolysis might be as influential in 22 generating volatile compounds as lipid oxidation. 23

Keywords: dry-cured loin, proteolysis, lipid oxidation, volatile compound, quality properties

26 Introduction

27 Lipid oxidation and proteolysis are factors that cause significant biochemical changes in 28 the quality characteristics of dry-cured meat. Previous studies have well-established the 29 relationship between quality characteristics and these factors (Toldrá, 1998). The majority of 30 volatile compounds originate from the chemical or enzymatic oxidation of unsaturated fatty 31 acids. These compounds can then further interact with proteins, peptides, and free amino acids. In 32 the case of lipid oxidation, fatty acids generated through lipolytic enzymes in dry-cured meat 33 generate volatiles through oxidation reactions (Toldrá, 2006; Gilles, 2009). Regarding 34 proteolysis, amino acids released from meat proteins form volatile compounds due to Strecker degradation and Maillard reaction (Luo et al., 2021; Pérez-Santaescolástica et al., 2018; Toldrá, 35 36 1998). Volatile compounds generated by these two biochemical pathways are a key factor in 37 aroma/flavor enhancement, and independent research on their effects on quality characteristics 38 has been conducted by researchers. Górska et al. (2017) divided 21 dry-cured loins into three 39 groups according to the characteristics of volatile compounds through principal component 40 analysis (PCA) and evaluated the correlation between volatile compounds and sensory 41 characteristics through sensory evaluation of each group. In addition, the relationship between 42 volatile compounds and sensory properties as well as the relationship between physicochemical 43 properties has been demonstrated (Marušić Radovčić et al., 2016; Petričević et al., 2018). Luo et 44 al. (2021) mentioned that 3-methylbutanal is a compound produced by the interaction between 45 leucine and some branched amino acids, and Villalobos-Delgado et al. (2014) reported that 46 benzene acetaldehyde is produced through Strecker degradation from phenylalanine. In addition, 47 it has been reported that furan and non-heterocyclic sulfur compounds are produced (Flores, 2018). Consequently, amino acids produced by proteolysis not only act as flavor precursors 48

49 themselves but also affect various flavor/aroma-related compounds through additional

biochemical reactions. Therefore, a better understanding of lipid oxidation and proteolysis during
manufacturing may contribute to generating the volatile compounds in final products.

52 Many previous studies have reported the effects of lipid oxidation or proteolysis levels on 53 changes in the quality characteristics of dry-cured meat. Pérez-Santaescolástica et al. (2018) 54 compared quality characteristics in three dry-cured ham groups (low, proteolysis index < 32%; medium, $32\% \le$ proteolysis index $\le 36\%$; high, proteolysis index > 36%) according to the 55 56 proteolysis level, and the authors reported that an increase in proteolysis led to a rise in texture 57 profile analysis (TPA) adhesiveness and a decrease in the quantity and quality of volatile compounds. They argued that the processing conditions conducive to lipid oxidation, such as 58 59 increased salt content, typically, restrict the function of proteolytic enzymes. On the other hand, Marušić Radovčić et al. (2021) identified that during the manufacturing period, the biceps 60 61 femoris, where proteolysis was high, exhibited a greater increase in volatile compounds than the 62 semimembranosus. In addition, the authors mentioned that a moderate degree of lipid oxidation 63 could have a positive effect on the flavor enhancement of dry-cured ham and that a high degree 64 of lipid oxidation caused damage to the nutritional, functional, and sensory properties. Previous 65 studies have shown that not only volatile substances but also other quality characteristics (texture 66 properties, color, fatty acids, amino acids, etc.) could be changed by lipid oxidation or proteolysis. 67

Although the effect of lipid oxidation or proteolysis on quality characteristics is also important, understanding the relationship between them is essential for a better understanding of the quality characteristics. However, research related to elucidating the relationship between lipid oxidation and proteolysis is very limited. A discussion on the relationship between lipid oxidation 72 and proteolysis can only be found in one paper. Harkouss et al. (2015) reported a quantitative 73 correlation between lipid oxidation and proteolysis and established a linear regression model with 74 a high coefficient of determination, regardless of muscle type, from 0 to 21 weeks during the entire manufacturing period of dry-cured ham ($R^2=0.92$). The scarcity of this information is 75 76 likely due to the inability in the manufacturing process to selectively initiate chemical reactions 77 associated solely with protein or lipid oxidation or degradation. To overcome this limitation, in 78 this study, the opposite of the experimental group was confirmed through a statistical procedure 79 for the study group based on the TBARS and proteolysis index of the generally prepared dry-80 cured loin. Furthermore, this study has the advantage of considering not only proteolysis but also 81 lipid oxidation at the same time.

82 Lipid oxidation and proteolysis-related changes are biochemical reactions that occur 83 simultaneously. Therefore, the relationship between the two reactions needs to be clearly 84 established. Our hypothesis suggests that proteolysis may substantially impact quality 85 characteristics considering the approximate composition of raw meat. Additionally, the effect of 86 lipid oxidation may be greater when it is a limited volatile compound. The purpose of this study 87 was to compare the quality characteristics of dry-cured loins by grouping them according to the 88 levels of lipid oxidation and proteolysis, and to discuss the relationship between lipid oxidation 89 and proteolysis on quality characteristics.

90

91 Material and Methods

92 Sample preparation

93 Sixty pork loins at 2 d postmortem from the right side of the carcass were employed for
94 this study. Pork loin was purchased six times at two-month intervals from a local butcher block in

95 Jinju, Korea. At the time of each purchasing, the pork loins were assigned a sample code (random 96 4-digit code). The coded pork loins were left in a cooler until 3 d postmortem before 97 manufacturing dry-cured loins. The fresh pork loin was salted after all connective tissues were 98 removed. For salting, purified sodium chloride was used and 4% of the pork loin weight was 99 rubbed on the surface. Then, it was vacuum-packed and stored in a refrigerator at 4°C for 7 d. 100 The surface of the cured loin was washed in tap water and the water on the surface was removed 101 with a paper towel. The cured loins were placed in a permeable fibrous casing and dried at 9°C 102 (50~55% RH) for 14 d, then the temperature was raised by 1°C every 3 d, and finally aged at 103 18°C (65~70% RH). The total production period was 55 d. The dry-cured loins were removed 104 from the casing and stored at -80°C until analysis. To assign samples based on lipid oxidation 105 and proteolysis, the dry-cured loins were measured for lipid oxidation using 2-thiobarbituric acid 106 reactive substances (TBARS) and for proteolysis using proteolysis index (PI), as described in Section 2.2. After dividing the samples into four groups based on TBARS and PI, detailed 107 108 information is presented in Table 1.

109

110 2-thiobarbituric acid reactive substances and proteolysis index

The TBARS of dry-cured loin was determined with Bozkurt and Erkmen (2004) with some modifications. A 3 g of sample was homogenized with 27 mL of 3.86% perchloric acid using homogenizer (T25 Basic, IKA, Wilmington, NC, USA). The homogenate stood in a cold room for an hour. After standing, the homogenate was centrifuged at 2000×g for 10 min and filtered with Whatman No. 1 filter paper. The 2 mL of filtrate was mixed with 2 mL of 20 mM TBA solution and left for 16 hours at room temperature with dark conditions. For the blank, 2 mL of deionized water was used to replace for filtrate. The TBARS was measured at 531 nm using Cary 60 spectrometer (Agilent Technologies, Santa Clara, CA, USA). A standard curve
was calculated using 1,1,3,3-tetraethoxypropane.

120 The total nitrogen content (TN) was measured with 0.5 g of sample using the nitrogen121 analyzer (TruMac CNS analyzer, Leco, MI, USA).

122 The non-protein nitrogen content was determined according to Pérez-Santaescolástica et 123 al. (2018) with some modifications. A 3 g minced sample was homogenized with 25 mL of 124 deionized water, and 12 mL of 20% trichloroacetic acid (TCA) were introduced into the 125 homogenate. The mixture was thoroughly agitated and then left to stabilize for an hour at room 126 temperature. Subsequently, it was centrifuged again at 2000×g for 10 min. Post centrifugation, the supernatant was filtered. A 15 mL of the filtrate was then employed to determine the nitrogen 127 128 content, using the same procedure as was used for determining total nitrogen content by Kjeldahl 129 method. Finally, the proteolytic index (PI) was calculated using the formula (non-protein 130 nitrogen/total nitrogen)×100, as described by Pérez-Santaescolástica et al. (2018). 131 132 *Moisture, protein, and fat content* 133 Moisture was performed by drying method the samples to a constant weight at 103°C 134 (method 934.01, AOAC, 2005). The total nitrogen content was determined as described in

section 2.2. The protein content was calculated by multiplying the total nitrogen content by 6.25.

136 The fat content was conducted with Soxhlet extraction method with diethyl ether (method

137 920.39, AOAC, 2005).

138

139 Instrumental color, pH, and texture profile analysis (TPA)

140 The dry-cured loins were cut to the same size [8 cm (W) \times 5 cm (L) \times 3 cm (H)], and three pieces were used for instrumental color. Each piece was measured at different locations five 141 142 times. The instrumental color was analyzed by colorimeter (CR-400, Konica Minolta, Tokyo, Japan) with an 8 mm aperture using an illuminant with D65 and 2° standard observer after 143 144 calibrating with a calibration plate (Y = 81.2; x = 0.3191; y = 0.3263). The Commission Internationale de l'Eclairage (CIE) L^* (lightness), a^* (redness), and b^* (yellowness) were used as 145 146 color parameters. The mean value was read 5 readings for each piece and took the average of 3 147 pieces.

The 3 g of minced sample was homogenized with 27 mL of deionized water using 148 149 homogenizer (T25 Basic, IKA). The pH of the sample was measured with a pH meter (S20 150 SevenEasyTM, Geifensee, Switzerland) after calibrating with 7.00, 4.01, and 9.21 using pH buffer. 151 The TPA was performed using EZ-SX (Shimadzu Corp., Kyoto, Japan) with a flat bottom 152 (5 mm diameter) probe. The 3 pieces of the sample was used after measuring the instrumental 153 color. A double compression test was carried out with 50% compression of the original sample 154 height. Force-time curves were expressed with a 500 N load cell applied at a 100 mm/min 155 crosshead speed. The texture parameters were calculated according to Bourne (1982). The mean 156 value was taken from two random points in each sample piece.

157

158 Free amino acid composition

Free amino acid composition was extracted as described by Aro et al. (2010) with some modifications. Briefly, a 3 g of sample and 27 mL of deionized water were homogenized for 30 s and 10 mL of 10% TCA solution was added and shaken at 250 rpm for 1 h at room temperature. After an hour, centrifugation was performed at 10,000×g for 10 min and the supernatant was 163 filtered using Whatman No.1. Finally, it was filtered with a 0.2 µm syringe filter, transferred to a

164 1.5 mL vial, and used for analysis. A 20 µL of sample was injected into amino acid analyzer

165 (Biochrom 30, Biochrom, Cambridge, UK). Amino acids were derivatizated with 6-

166 aminoquinolyl-Nhydroxysuccinimidyl carbamate (Waters AccQ-Fluor reagent kit) and used the

167 standard with 40 amino acids (Physiology standard, Biochrom, Cambridge, UK). The results

168 were expressed as mg of free amino acid/100 g of sample.

169

170 *Lipid extract and fatty acid profile*

171 Lipid extraction was carried out according to Folch et al. (1957). Saponification,

172 methylation, and gas chromatography learning conditions were employed by Seo et al. (2021).

173 The fatty acid methyl etsers (FAME) were separated with gas chromatography (7890B, Agilent

174 Technologies, CA, USA) equipped with a Supelco SP[®] -2560 capillary gas chromatography (GC)

175 column (100 m \times 0.25 mm \times 0.20 μ m). The individual FAMEs were identified based on a

176 standard mixture of Supelco 37 Component FAME Mix (Sigma Aldrich, MO, USA). The amount

177 of individual fatty acids and fatty acid groups identified was calculated with internal standard

178 (C11:0, undecanoic acid). The fatty acid was expressed as g/100 g tissue.

179

180 Volatile compound extraction and injection

181 The SPME fiber was used after pre-conditioning at 270°C for 30 min according to the

182 manufacturer's manual before use. Briefly, 2.5 g of the minced sample was placed in a 20 mL

- 183 screw-capped vial and sealed with a screw cap fitted with PTFE septa. The adsorption of volatiles
- 184 was carried out for 55 min in an incubator at 45°C using 50/30 µm DVB/CAR/PDMS SPME
- 185 fiber (57328-U, Supelco, PA, USA). After adsorption, the SPME fiber was immediately injected

into the GC inlet and desorbed for 3 min in pressure splitless mode (30 psi). At this time, the GC
inlet temperature was 240°C. Before adsorbing the next sample, pre-conditioning was performed
for 8 min while flowing carrier gas (Helium, 99.999%) at 240°C at 100 mL/min for overall
SPME fiber cleaning.

190

191 GC-mass spectrometry

192 The GC machine was 7890B (Agilent Technologies, CA, USA). Helium was used as the 193 carrier gas and a constant flow rate of 1.0 mL/min was used. The GC column was a Supelcowax 194 10 capillary column (60 m \times 0.32 mm \times 0.25 µm, Supelco, Bellefonte, PA, USA). The oven 195 temperature program is described in Domínguez et al. (2019) and analyzed for a total of 49.5 196 min. A mass spectrometer of 5975C (Agilent Technologies, CA, USA) was used, and the 197 temperatures of the mass transfer line, ion source, and quadrupole were 260°C, 230°C, and 198 150°C, respectively. Mass spectra were obtained at 70 eV electric shock and EMV of 1080, and 199 data were collected in the m/z range of 40-450 at a scan speed of 3.4 times per second. 200 Compounds were identified by comparing their mass spectra with those contained in the NIST 201 20/Wiley 12th library. The results are expressed as area units (AU) \times 10⁵/g of sample. 202

203 Statistical analysis

All quality characteristics were taken an average of three replication on each measurement for further statistical analysis. The distribution of sample groups was expressed as a score plot using principal component analysis using data from physicochemical variables by MetaboAnalyst (version 5.0; https://www.metaboanalyst.ca/). Mixed model was used to confirm the difference in physicochemical properties in the sample groups according to the TBARS value and PI. The mixed model was performed using the mixed procedure of SAS 9.4. The statistical
models were used to evaluate the quality characteristics (moisture, instrumental color, pH, texture
profile analysis, fatty acid content, free amino acid content, and volatile compounds) of the drycured loin. These models included a fixed effect for the sample group and a random terms for
processing batch and sample replication. Least squares means for all traits of interest were
separated (F test, p<0.05) by using least significant differences analyzed by the PDIFF option in
SAS.

217

218 **Results and Discussion**

219 Sample construction

220 Table 1 and Fig. 1 show the sample descriptions used in this study. PI and TBARS have been widely used as indicators of proteolysis and lipid oxidation in many previous studies, 221 222 respectively (Marušić Radovčić et al., 2021). The sample groups used in this study showed 223 significant differences in proteolysis and lipid oxidation levels (p < 0.05). Sample grouping was 224 also performed using principal component analysis (PCA) (Fig. 2). As a result of the PCA, PC1 225 and 2 from sample groups were explained 73% variance, which may consider a clear distribution 226 between the groups. Therefore, the grouping of samples based on PI and TBARS appears to be 227 highly appropriate.

In dry-cured ham, the PI reflecting excellent quality was reported in previous studies as 33-36% (Spanish style) and 22-30% (Italian style) (Careri et al., 1993; Pérez-Santaescolástica et al., 2018). However, in products using pork loin, there is no mention of the PI range for good quality yet. The PI values in the current study ranged from 7% to 14%, consistent with data 232 reported by other authors (Abellán et al., 2018; Stadnik & Dolatowski, 2013). This difference 233 between dry-cured ham and dry-cured loin is due to manufacturing conditions such as raw 234 material, salting period, and drying-ripening period. Our TBARS (0.6~1.1 mg MDA/kg sample) 235 results are consistent with those reported by Muriel et al. (2007) and Soto et al. (2008). In 236 addition, Hoz et al. (2007) reported that the overall acceptability of dry-cured loins with 0.87 to 237 1.24 mg of TBARS was higher than that of dry-cured loins with higher TBARS. The results 238 suggest that the dry-cured loin utilized in this study can be characterized as exhibiting normal 239 quality. Therefore, despite the absence of a definitive statement regarding the specific ranges of 240 PI and TBARS, dividing these values into four groups based on the median value from the 241 present study may be a somewhat compelling approach.

242

243 *Physicochemical properties*

Table 2 presents the results of physicochemical properties for the four dry-cured loin 244 245 groups based on the levels of proteolysis and lipid oxidation. Moisture content exhibited a 246 significant difference between groups (p<0.05), with the HH group showing the highest moisture 247 content at around 51%, and the LL group exhibiting the lowest at around 47%. In protein content, 248 on the other hand, high proteolysis groups (HH and HL) were significantly lower than low 249 proteolysis groups (p < 0.05). Also, the fat content was observed same significant result as for 250 protein content (p<0.05). Redness showed a significant difference among the sample groups, with 251 HH and HL exhibiting higher values, while LH and LL showed relatively lower values (p < 0.05). 252 TPA hardness and cohesiveness showed significant differences across groups (p<0.05). HH and 253 HL exhibited lower TPA hardness than LH and LL, while TPA cohesiveness was higher. Thus, 254 the level of proteolysis and lipid oxidation did not significantly affect the lightness, yellowness,

and pH (p>0.05), but they notably influenced the other variables, with proteolysis seen to giving
a more substantial effect than lipid oxidation.

257 The current results of moisture, protein and lipid content in dry-cured loin were agree 258 with previous studies (Bermúdez et al., 2015; Ortiz et al., 2020; Seong et al., 2015). Moisture 259 content is an important factor in determining the quality of dry-cured meat products, as the level 260 of moisture dehydration directly affects tissue properties and various enzyme activities (Harkouss 261 et al., 2015). In addition, moisture is one of the technological traits that determine not only 262 dryness but also storage stability of dry-cured meat (Seong et al., 2015). HH exhibited a higher 263 moisture content than HL, indicating weaker dehydration in HH than in HL. Accordingly, lipid oxidation and proteolysis were more active in HH than in HL. Moreover, the difference in 264 265 moisture content between HH and HL was only about 1%, whereas HH and LH showed a 266 difference of about 3%. Therefore, proteolysis may have a more significant influence than lipid 267 oxidation on moisture, protein and fat content in dry-cured loin. Additionally, both the protein 268 and fat contents changed due to variations in moisture content brought about by dehydration. In 269 particular, the protein content increased, reaching levels similar to that of the water content. Such 270 changes can significantly impact subsequent biochemical reactions, potentially altering the 271 reaction products.

Our findings align with Pérez-Alvarez et al. (1999) report of high *a** values in dry-cured ham muscles with high moisture content, citing nitrosomyoglobin formation as the cause. The a* value may be influenced by the denaturation of nitrosomyoglobin due to the production of lactic acid during ripening (Lorenzo et al., 2013). Therefore, it can be hypothesized that proteolysis affected nitrosomyoglobin and production of lactic acid. However, this explanation alone does not sufficiently clarify why the sample group with higher proteolysis exhibited stronger redness. Thus, additional research is needed on the relationship between proteolysis and color in dry-curedloin.

280 The observed hardness and cohesiveness cannot be entirely explained by moisture content 281 alone, as proteolysis and lipid oxidation effects must also be considered. In HH and HL, the dry 282 matter's interference in TPA measurement is reduced due to high moisture content, while LH and 283 LL, with low moisture content, increase the interference of dry matter, resulting in the current 284 hardness result. Dry-cured loin shrinks gradually during manufacturing, which may result in 285 increased cohesiveness. LH and LL, which were highly dehydrated, contracted more strongly, 286 and cohesiveness increased as the dehydrated empty space naturally decreased. Seo et al. (2021) 287 also reported similar findings to ours.

288

289 *Fatty acid profile*

290 Table 3 shows the fatty acid content in relation to the levels of proteolysis and lipid 291 oxidation. No significant difference was observed in saturated fatty acids (SFA) between the 292 sample groups (p>0.05). However, monounsaturated fatty acids (MUFA) and polyunsaturated 293 fatty acids (PUFA) exhibited significant differences between the groups with strong significance 294 levels of p<0.05 and p<0.001, respectively. Concerning individual fatty acids, significant 295 differences were found between sample groups for SFA at C10:0, C12:0, and C20:0 (p<0.05). 296 For MUFA and PUFA, there were no significant differences between sample groups in C18:1 n-7 297 and C20:4, respectively, but significant differences were observed in other fatty acids (p < 0.05). 298 In particular, there was a significant difference for C16:1, C18:2 n-3, C18:3 n-6, C20:1, C20:2 299 and C20:3 (p < 0.05). The amount of MUFA was significantly lower in HH than other sample 300 groups (p<0.05). On the other hand, there was no significant difference between LH and LL in

301 MUFA(p>0.05). Therefore, MUFA was affected by lipid oxidation in the high proteolytic sample 302 group but not affected by lipid oxidation in the opposite case. In terms of the PUFA, HH and HL 303 were significantly lower than LH and LL (P<0.05). These results indicated that higher proteolytic 304 levels were associated with lower amount of PUFA, regardless of the degree of lipid oxidation. 305 Conversely, lower proteolytic levels were linked to higher amount of PUFA, regardless of the 306 degree of lipid oxidation. Additionally, there does not appear to be a noticeable effect of lipid 307 oxidation on changes in PUFA. These results were reflected in the total fatty acid content, and 308 HH and HL had a significantly lower total fatty acid content than LH and LL (p<0.05). 309 Therefore, it may be suggested that proteolysis contributes more to changes in fatty acid content 310 than lipid oxidation. 311 Previous studies have reported that the content of SFA, MUFA, and PUFA decreases in the final product compared to the initial stage (Gilles, 2009; Andres et al., 2005; Salazar et al., 312 313 2016). During this time, mainly neutral lipids and phospholipids decrease (Gilles, 2009; Andres 314 et al., 2005; Salazar et al., 2016). In the present study, proteolysis induced more changes in fatty 315 acid content than lipid oxidation, and these changes are mainly attributable to PUFAs. The most 316 probable hypothesis that can be inferred is the interaction between protein and lipid oxidation. 317 Oxidation reactions may easily lead from fat and protein, causing various changes in food. 318 Among these changes, the loss of enzyme activity, which can directly affect the quality of dry-319 cured meat products, is a representative example (Viljanen et al., 2004). Considering that our dry-320 cured loin had a significantly higher protein content than fat content on a dry matter basis (Table

321 2) and that the oxidation reaction is a chain reaction, the following conclusions can be drawn.

322 Fat-initiated oxidation can induce a cascade of oxidation reactions and promote the oxidation of

323 proteins. Carbonylated proteins may be more susceptible to enzyme-induced proteolysis

324 (Nyström, 2005). Gan et al. (2019) also mentioned the relationship between proteolysis and 325 protein oxidation, arguing that protein oxidation can cause changes to protein structure, leading to 326 unfolded proteins that are more reactive forms for proteases. Consequently, the final products 327 such as hydroperoxide from protein-lipid-derived oxidation may have induced changes in fatty 328 acids (Viljanen et al., 2004; Lund et al., 2011). Among fatty acids, PUFA is known to undergo 329 the most severe changes due to oxidation, with n-6 type fatty acids being the most severely 330 reduced as oxidation proceeds (Sampaio et al., 2012). Therefore, in the group with high 331 proteolysis, the content of PUFAs may have decreased, which may have affected the volatile 332 compounds.

333

334 Free amino acid

335 Table 4 presents the content of free amino acids. Except for taurine and anserine, there was a significant difference between groups overall (p<0.001). HH had the highest content at 336 337 approximately 2309 mg and LL had the lowest content at 1799 mg (p<0.05). Furthermore, there 338 was no significant difference between HL and LH, with both having a content between 1880 and 339 1895 mg (p<0.05). Aspartic acid, threonine, alanine, valine, and isoleucine showed the same 340 results among sample groups, with HH having the highest content and LL having the lowest 341 content (p<0.05). Similarly, serine, asparagine, glutamic acid, glycine, isoleucine, leucine, 342 tyrosine, and phenylalanine showed the same trend, with the highest content in HH and the 343 lowest content in LL (p<0.05). Additionally, there was no significant difference between HL and 344 LH (p<0.05). These findings were consistent with those of Cittadini et al. (2020), who reported 345 that free amino acid content was high in the group with high TBARS. In summary, proteolysis or 346 lipid oxidation could influence the generation of free amino acids, with high levels of either

leading to an increase in the content of free amino acids. Furthermore, high levels of both couldlead to the highest content of free amino acids.

349 As discussed earlier, free amino acids are considered as representative indicators of 350 proteolysis. Therefore, the changes in free amino acid levels according to various conditions have 351 been investigated in several previous studies. Garrido et al. (2012) reported that the total amino 352 acid content decreased with an increase in salting time, as higher salt concentrations can inhibit proteolytic activity. Protein degradation through cathepsin B, L, and calpain activity promotes the 353 354 production of small peptides and free amino acids. In particular, myosin heavy chain, C protein, 355 and desmin are degraded by proteolytic enzymes, which can contribute to the generation of free 356 amino acids (Zhou et al., 2017).

357 Lipid oxidation is a significant chemical reaction that occurs in meat and meat products. It 358 is a breakdown of lipid components and is closely related to lipolytic activity, which releases free 359 fatty acids (Muriel et al., 2007). Reactive oxygen species typically initiate lipid oxidation by 360 reacting with fatty acids (Domínguez et al., 2019). Lipid oxidation products such as hydrogen 361 peroxide, lipid-free radicals, and volatile secondary oxidation products can react with proteins to 362 generate protein-centric free radicals (Hematyar et al., 2019). Furthermore, carbonylation of 363 proteins is promoted by lipid oxidation, and carbonylated proteins are more sensitive to 364 proteolytic enzymes (Nyström, 2005). Moreover, the folded structure of natural proteins forms a 365 rigid structure that makes it difficult for proteolytic enzymes to act as a substrate (Fontana et al., 366 2004). However, oxidation-induced changes in the secondary and tertiary structures of proteins 367 result in an unfolded state, leading to an increase in the site of action of proteases and protein 368 degradation (Grune et al., 2004). This ultimately leads to an increase in the production of free 369 amino acids released by proteolysis.

3.5. Volatile compound

372	The volatile compounds were analyzed by the sample group and shown in Table 5. A total
373	of 27 volatile compounds were detected, consisting of 11 alcohols, 4 alkanes, 6 ketones, and 6
374	aldehydes. Of these, 7 alcohols, 3 alkanes, 3 ketones, and 4 aldehydes showed significant
375	differences between groups (p<0.05). For \sum alcohol, \sum alkane, \sum ketone, and \sum total, HH and HL
376	were high, while LH and LL were low (p<0.05). However, \sum aldehyde showed the opposite result
377	(p<0.05). Therefore, the sample group with high protein degradation (HH and HL) produced a
378	large amount of total volatile compounds. Interestingly, there was no difference according to lipid
379	oxidation in the sample group with high protein degradation, but there was a significant
380	difference according to lipid oxidation in the opposite case ($p < 0.05$).
381	An additional discussion was not conducted because the properties of individual volatile
382	compounds are widely known in dry-cured meat.
383	The formation of volatile compounds in dry-cured meat products is primarily considered a
384	result of lipid oxidation. Due to the lipolytic activity (enzymatic and oxidative) of dry-cured
385	meat, unsaturated fatty acids are converted into various compounds, mainly aliphatic compounds
386	such as alcohols, aldehydes, and ketones. In dry-cured meat, volatiles are mainly produced by
387	lipid oxidation in the initial processing step, whereas in the ripening step, both lipid and amino
388	acid degradation occur (Gilles, 2009). In fact, among the results of our study, the amount of the
389	PUFA and volatile compounds showed the same manner between groups, which is likely due to
390	the effect of proteolysis. Therefore, the change in PUFA was caused by proteolysis, and it could
391	be inferred that the current volatile compound result is also due to this. Earlier, it was explained
392	that oxidation was carried out by proteolysis and had an effect on PUFA. A previous study

393 reported that lipid oxidation is closely related to volatile compounds (Domínguez et al., 2014). 394 Additionally, lipid oxidation usually occurs in polyunsaturated fatty acids, as the double bond site 395 has a radical reactive site, making it more sensitive with more double bonds (Domínguez et al., 396 2019). Therefore, during the dry-aging process of dry-cured loin, polyunsaturated fatty acids are 397 converted to volatile compounds due to lipid oxidation, generating alcohols, aldehydes, and 398 ketones. Moreover, free amino acids are known to act as precursors of volatile compounds and 399 are mainly produced by Strecker degradation and Maillard reaction, with heterocyclic compounds 400 containing sulfur or nitrogen and aliphatic compounds being produced (Toldrá, 1998; Jurado et 401 al., 2007).

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- 403

404 Conclusion

The study investigated how proteolysis and lipid oxidation affect key quality parameters 405 406 during the manufacturing of dry-cured loin. The results of this study demonstrated that varying 407 levels of proteolysis and lipid oxidation can influence the physicochemical properties, total fatty 408 acid content, total free amino acids content, and amount of volatile compounds in dry-cured loin. 409 While lipid oxidation did not affect the fatty acid content, it did influence the free amino acid 410 content between sample groups. Proteolysis, however, impacted both. Intriguingly, volatile 411 compounds were influenced by the level of lipid oxidation when proteolysis was high but 412 remained unaffected by lipid oxidation levels when proteolysis was low. These results implicate 413 that lipid oxidation and proteolysis can directly cause changes in quality characteristics when 414 manufacturing dry-cured loin. It also suggests that not only lipid oxidation but also proteolysis 415 may play an important role in the formation of volatile compounds in dry-cured loin.

417

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552 Figure legends

- 553 Fig. 1. The values of PI (upper) and TBARS (below) for the sample group. HH, PI and TBARS
- of high levels; HL, PI of high level and TBARS of low level; LH, PI of low level and TBARS of
- high level; LL, PI and TBARS of low levels. Error bar represents the standard error. ^{a-d}Different
- small letters within different sample groups were significantly different (p<0.05).
- 557
- 558 Fig. 2. The score plot was expressed to sample distribution in one or two PC. HH, PI and TBARS
- of high levels; HL, PI of high level and TBARS of low level; LH, PI of low level and TBARS of
- 560 high level; LL, PI and TBARS of low levels.
- 561

Sample classification ¹	PI range	TBARS range	n
НН	<i>X</i> ≥ 10.6	$X \ge 0.88$	16
HL	$X \ge 10.6$	X < 0.88	9
LH	<i>X</i> < 10.6	$X \ge 0.88$	18
LL	<i>X</i> < 10.6	X < 0.88	17

562 Table 1. Sample classification of dry-cured loin by PI and TBARS

⁵⁶³ ¹HH, PI and TBARS of high levels; HL, PI of high level and TBARS of low level; LH, PI of low level and

564 TBARS of high level; LL, PI and TBARS of low levels.

Item	HH^1	HL	LH	LL	SEM ²	p-value ³
Moisture (%)	50.87a	49.54b	47.59c	46.65d	0.52	***
Protein (%)	42.04c	42.71b	43.69ab	43.88a	0.33	**
Fat (%)	5.59b	6.15b	7.02a	7.87a	0.10	*
Lightness	44.51	45.26	46.67	44.20	0.72	ns
Redness	7.63a	7.56a	6.57b	6.83b	0.27	*
Yellowness	5.11	5.66	5.82	5.61	0.24	ns
pH	5.96	5.99	6.01	6.07	0.03	ns
Hardness (N)	19.07b	18.81b	20.95a	21.23a	0.66	**
Cohesiveness	0.46a	0.46a	0.42b	0.42b	0.01	***

566 Table 2. Physicochemical properties for sample groups determined by PI and TBARS

⁵⁶⁷ ¹HH, PI and TBARS of high levels; HL, PI of high level and TBARS of low level; LH, PI of low level and

568 TBARS of high level; LL, PI and TBARS of low levels.

569 ²Standard error of the means.

570 ³*, p<0.05; **, p<0.01; ***, p<0.001.

571 a-dDifferent small letters within different sample groups were significantly different (p<0.05).

572 ns, no significance.

Fatty acid content	HH^{1}	Ш	ТН	TT	SEM ²	n value ³	
(mg/100g tissue)	1111	IIL	LII	LL	SEM	p-value	
C10:0	28.33c	29.08c	33.92a	30.85b	0.98	*	
C12:0	29.22c	31.80b	34.86a	33.65a	1.10	*	
C14:0	154.06	169.91	162.98	167.31	11.57	ns	
C16:0	2156.83	2208.81	2222.41	2202.15	20.29	ns	
C16:1	330.25a	323.46a	331.62a	307.52b	9.86	*	
C17:0	56.67	46.34	56.53	47.67	7.43	ns	
C18:0	1100.55	1090.32	1124.87	1119.77	96.53	ns	
C18:1 n-7	358.59	352.54	374.96	352.38	26.61	ns	
C18:1 n-9	3493.79b	3572.62a	3629.91a	3639.72a	44.48	*	
C18:2 n-3	42.50c	55.42b	64.06a	65.43a	3.52	***	
C18:3 n-6	751.70c	846.82b	976.96a	979.57a	35.40	***	
C20:0	31.87c	34.53b	39.57a	37.39a	1.62	**	
C20:1	79.69b	89.04a	90.44a	90.67a	5.55	**	
C20:2	43.38c	49.97b	55.58ab	57.95a	3.17	***	
C20:3	38.96b	39.98b	48.99a	43.93b	2.23	*	
C20:4	158.49	152.64	175.23	172.92	17.35	ns	
SFA	3556.65	3611.69	3674.19	3636.92	56.91	ns	
MUFA	4262.32b	4329.48a	4426.93a	4391.22a	67.95	*	
PUFA	1035.03b	1144.84b	1320.82a	1318.86a	74.43	***	
n-6/n-3	17.69	15.28	15.25	14.97	2.03	ns	
Total	8854.00b	9086.00b	9421.00a	9347.00a	104.41	*	

574 Table 3. Fatty acid content for sample groups determined by PI and TBARS

¹HH, PI and TBARS of high levels; HL, PI of high level and TBARS of low level; LH, PI of low level and

576 TBARS of high level; LL, PI and TBARS of low levels.

577 ²Standard error of the means.

578 ³*, p<0.05; **, p<0.01; ***, p<0.001.

579 a-cDifferent small letters within different sample groups were significantly different (p<0.05).

580 ns, no significance.

581 SFA, saturated fatty acid; MUFA, monounsaturated fatty acid; PUFA, polyunsaturated fatty acid.

Free amino acid	HH^{1}	НІ	ТН	II	SEM ²	n-value ³
(mg/100g dry-cured loin)	1111	IIL	LII	LL	SLIVI	p-value
Taurine	105.45	103.70	87.96	89.94	5.89	ns
Aspartic	62.63a	42.11c	51.20b	42.73c	3.28	***
Threonine	130.14a	99.53c	102.69b	95.00c	4.84	***
Serine	77.63a	59.16b	61.28b	55.62c	2.76	***
Asparagine	87.54a	67.37b	69.55b	62.26c	2.95	***
Glutamic	248.69a	200.99b	198.03b	189.92c	7.96	***
Proline	71.22	63.36	60.74	69.51	5.49	ns
Glycine	118.05a	94.95b	94.42b	86.23c	4.20	***
Alanine	214.74a	164.56c	173.92bc	167.88c	6.74	***
Valine	132.67a	105.96c	114.05b	109.5c	4.98	**
Methionine	72.91a	61.71b	59.43c	54.71d	1.93	***
Isoleucine	104.36a	85.84bc	89.01b	84.76c	3.34	***
Leucine	175.04a	143.35b	145.22b	135.52c	5.46	***
Tyrosine	103.78a	85.95b	86.51b	81.75c	2.87	***
Phenylalanine	87.37a	75.42b	73.90b	68.82c	2.23	***
Ornithine	5.29b	4.06b	7.74a	6.29a	0.72	**
Lysine	246.63a	204.34b	209.20b	200.70b	8.55	***
Histidine	60.24a	53.74b	49.76c	48.47c	2.31	**
Tryptophan	16.18a	8.82b	6.57c	3.27d	1.65	***
Anserine	33.21	32.47	31.04	31.50	0.82	ns
Arginine	152.19a	122.13b	122.33b	114.75c	4.72	***
Total	2309.72a	1880.55b	1895.55b	1799.26c	78.51	***

583 Table 4. Free amino acid content for sample groups determined by PI and TBARS

¹HH, PI and TBARS of high levels; HL, PI of high level and TBARS of low level; LH, PI of low level and

585 TBARS of high level; LL, PI and TBARS of low levels.

586 ²Standard error of the means.

587 ³*, p<0.05; **, p<0.01; ***, p<0.001.

588 a-dDifferent small letters within different sample groups were significantly different (p<0.05).

589 ns, no significance.

590 TFAA, total free amino acid.

Compounds (AU×10 ⁵)	HH^{1}	HL	LH	LL	SEM ²	p-value ³
Heptanol	5.72a	5.85a	4.38b	4.59b	0.24	***
Hexanol	31.36a	25.58b	18.65c	16.29c	1.57	***
Nonanol	0.65	1.76	2.21	1.16	0.53	ns
Octanol	5.63b	5.75b	6.30ab	7.07a	0.62	*
1-Octen-3-ol	72.78	86.01	84.78	76.82	3.98	ns
Pentanol	38.99a	34.28b	27.99c	26.86c	1.83	***
3-Pentenol	0.43	0.54	0.40	0.37	0.04	ns
2,3-Butanediol	0.23	0.36	0.02	0.06	0.10	ns
3-Methyl-1-butanol	18.49a	19.64a	8.47b	7.15c	1.27	***
Benzeneethanol	5.95a	1.80b	1.82b	1.33b	0.60	***
Ethanol	29.07d	30.33c	31.97b	33.32a	0.77	***
Alcohol	209.30a	211.90a	186.70b	175.02c	6.00	***
2,3,5,8-Tetramethyldecane	1.82	1.43	1.53	1.54	0.23	ns
Decane	7.56a	7.20a	5.99b	5.92b	0.21	***
Dodecane	7.58a	7.20a	6.13b	5.64c	0.20	***
Tetradecane	1.55a	1.42a	0.68b	0.83b	0.17	*
Alkane	18.51a	17.25a	14.38b	13.93b	0.62	***
2,3-Octanedione	2.45	2.56	2.51	2.00	0.25	ns
2-Heptanone	5.48a	5.66a	5.40a	2.97b	0.58	**
2-Octanone	0.38	0.43	0.47	0.45	0.03	ns
2-Propanone	15.27a	11.96b	10.21c	9.35d	0.61	***
3-Octanone	1.61	2.22	0.93	1.30	0.38	ns
3-Hydroxy-2-butanone	3.01a	2.92a	1.75b	1.77b	0.19	***
Ketone	28.20a	25.74b	21.26c	17.84d	1.09	***
Benzaldehyde	3.38a	3.67a	1.89b	2.37b	0.44	*
Decanal	1.58	1.42	1.08	1.31	0.24	ns
Heptanal	8.55c	10.32b	10.17b	12.03a	0.61	***
Hexanal	24.17a	22.72b	21.32c	20.96c	0.67	**
Nonanal	33.39b	32.28b	38.84a	38.55a	1.77	*
Octanal	9.15	9.72	10.18	9.84	0.32	ns
Aldehyde	80.22	80.12	83.48	85.06	1.91	ns
Total	336.24a	335.01a	305.82b	291.85c	7.06	***

592 Table 5. Volatile compound for sample groups determined by PI and TBARS

⁵⁹³ ¹HH, PI and TBARS of high levels; HL, PI of high level and TBARS of low level; LH, PI of low level and

594 TBARS of high level; LL, PI and TBARS of low levels.

⁵⁹⁵ ²Standard error of the means.

596 ³*, p<0.05; **, p<0.01; ***, p<0.001.

597 a-dDifferent small letters within different sample groups were significantly different (p<0.05).

598 ns, no significance.



Fig. 1.





Fig. 2.