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**- Food Science of Animal Resources -**  
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ARTICLE INFORMATION	Fill in information in each box below
<b>Article Type</b>	Research Article
<b>Article Title</b>	Effects of loquat ( <i>Eriobotrya japonica</i> Lindl.) leaf extract with or without ascorbic acid on the quality characteristics of semi-dried restructured jerky during storage
<b>Running Title (within 10 words)</b>	Jerky with loquat leaf extract and ascorbic acid
<b>Author</b>	Se-Myung Kim, Tae-Kyung Kim, Min-Cheol Kang, Ji Yoon Cha, Hae In Yong, Yun-Sang Choi
<b>Affiliation</b>	Research Group of Food Processing, Korea Food Research Institute, Wanju 55365, Republic of Korea
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<b>ORCID (All authors must have ORCID) <a href="https://orcid.org">https://orcid.org</a></b>	Se-Myung Kim ( <a href="https://orcid.org/0000-0003-2250-7243">orcid.org/ 0000-0003-2250-7243</a> ) Tae-Kyung Kim ( <a href="https://orcid.org/0000-0002-6349-4314">orcid.org/ 0000-0002-6349-4314</a> ) Min-Cheol Kang ( <a href="https://orcid.org/0000-0002-9658-9045">orcid.org/ 0000-0002-9658-9045</a> ) Ji Yoon Cha ( <a href="https://orcid.org/0000-0002-1694-4343">orcid.org/ 0000-0002-1694-4343</a> ) Hae In Yong ( <a href="https://orcid.org/0000-0003-0970-4496">orcid.org/ 0000-0003-0970-4496</a> ) Yun-Sang Choi ( <a href="https://orcid.org/0000-0001-8060-6237">orcid.org/0000-0001-8060-6237</a> )
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**CORRESPONDING AUTHOR CONTACT INFORMATION**

For the <u>corresponding</u> author (responsible for correspondence, proofreading, and reprints)	Fill in information in each box below
First name, middle initial, last name	Yun-Sang, Choi
Email address – this is where your proofs will be sent	kcys0517@kfri.re.kr
Secondary Email address	
Postal address	Korea Food Research Institute, Wanju 55365, Korea

Cell phone number	
Office phone number	82-63-219-9387
Fax number	82-63-219-9076

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ACCEPTED

9           **Effects of loquat (*Eriobotrya japonica* Lindl.) leaf extract powder with or without**  
10           **ascorbic acid on the quality characteristics of semi-dried restructured jerky during**  
11   **storage**

13   Abstract

14           Deterioration of jerky during storage is a major concern; this is usually combated with  
15           natural or synthetic antioxidants. This study aimed to evaluate the quality characteristics of  
16           semi-dried restructured jerky with and without loquat leaf extract powder (LE) and ascorbic  
17           acid (AA) during storage for 180 days. The jerkies were formulated with 0, 0.15, and 0.3% LE  
18           and/or 0.05% AA (Control, no antioxidant; AA, 0.05% ascorbic acid; LE 0.15, 0.15% loquat  
19           leaf extract; LE 0.15-AA, 0.15% loquat leaf extract + 0.05% ascorbic acid; LE 0.3, 0.3% loquat  
20           leaf extract; LE0.3-AA, 0.3% loquat leaf extract + 0.05% ascorbic acid). LE is a phenolic  
21           compound, whose 1,1-diphenyl-2-picrylhydrazyl radical scavenging activity and metal  
22           chelating activity were found to be higher than AA. All antioxidant combinations having higher  
23           LE concentration and containing AA were effective in delaying protein and lipid oxidation  
24           compared to the control or AA. At the end of storage period, LE 0.15-AA and AA had higher  
25           CIE a\* value and lower shear force than the control. Therefore, the combination of 0.15% LE  
26           and 0.05% AA can result in reduced protein and lipid oxidation without any negative effect on  
27           the quality characteristics of semi-dried restructured jerky.

28  
29           Key words: semi-dried, restructured jerky, loquat leaf, ascorbic acid, sulfhydryl concentration

## 32 **Introduction**

33 Jerky is a traditional dried meat product preserved by drying to reduce water activity; it is  
34 convenient, has a rich nutrient content, and is shelf-stable without refrigeration (Coradini et al.,  
35 2019). Jerky needs to be dried to a water activity value of  $\leq 0.85$  to achieve stability  
36 (Triyannanto & Lee, 2016). However, the drying process leads to a tough texture of jerky (An  
37 et al., 2010). While semi-dried or restructured jerky has a softer texture, it has high water  
38 activity resulting in protein and lipid oxidation (Yang et al., 2009), which is a major cause of  
39 deterioration of jerky quality during storage (Kim et al., 2022; Wongwiwat & Wattanachant,  
40 2015).

41 Control or minimization of protein and lipid oxidation during storage of meat products can  
42 be accomplished using synthetic or natural antioxidants (Kim et al., 2014). However, the safety  
43 of synthetic antioxidants, such as 2,6-di-*t*-butyl-*p*-hydroxytoluene (BHT) and *t*-butyl-4-  
44 hydroxyanisole (BHA), is a concern, with respect to consumer health, due to their potential  
45 toxicological effects (Nassu et al., 2003). Therefore, demand for natural antioxidants has  
46 increased in the recent years (Xu et al., 2018). Use of natural antioxidants is considered safe  
47 and is readily accepted by consumers; moreover, the legislation does not require safety tests in  
48 case of “generally recognized as safe (GRAS)” (Nassu et al., 2003).

49 Ascorbic acid (AA) has been widely accepted to improve storage stability of meat products  
50 (Haak et al., 2009). It is approved as a GRAS substance and acts as a synergist when applied  
51 in combination with other antioxidants, promoting their antioxidant activity as well (Hwang et  
52 al., 2013). Li et al. (2013) had shown that plant polyphenols in combination with ascorbic acid  
53 delayed lipid oxidation of dry-cured sausages, thereby maintaining their storage quality.  
54 Hwang et al. (2013) had reported that a combination of *ganghwayaksuk* extract and ascorbic  
55 acid delayed lipid oxidation in raw chicken patties during storage, hence extending their shelf  
56 life.

57 Loquat (*Eriobotrya japonica* Lindl.) is a medicinal plant belonging to the Rosaceae family  
58 (Fu et al., 2019). It is commercially cultivated in Korea, India, Italy, and many other countries,  
59 and loquat leaf can be harvested regardless of the season (Dhiman et al., 2021).  
60 Pharmacological studies have shown that loquat leaf extract powders (LE) contain abundant  
61 polyphenols, including ellagic acid, chlorogenic acid, and neochlorogenic acid, which have  
62 anti-oxidant and anti-inflammatory effects (Kim et al., 2019). Till date, some studies applied  
63 loquat leaf to foods such as dumpling or fish cake (Park, 2012; Park 2014). However, there has  
64 been no studies about meat product using loquat leaf, despite the high antioxidant capacity of  
65 loquat leaf. Accordingly, this study hypothesized that using loquat leaves has the potential to  
66 delay lipid oxidation values in meat products, and which could be further improved when  
67 combined with ascorbic acid.

68 Thus, the objective of this study was to investigate the synergistic effects of loquat leaf  
69 extract powder and ascorbic acid on quality characteristics of semi-dried restructured jerky  
70 during storage at room temperature.

71

## 72 **Materials and Methods**

### 73 *Preparation of loquat (*E. japonica* Lindl.) leaf extract powder*

74 To prepare loquat leaf extract powder (LE), loquat leaf was purchased from Handsherb  
75 (Yeongcheon, Korea). The leaves were rinsed and dried in shade. The dried leaves were  
76 extracted with 80% ethyl alcohol at 60 °C for 3 h and the solvent was evaporated at 45 °C for  
77 48 min in a vacuum extractor-concentrator (SSEE-1, SSE, Bucheon, Korea). The extract was  
78 then stored at -45 °C for 10 h at 2 mTorr and made into a powder using a freeze dryer (VTFD,  
79 Ilshin, Korea). The same procedure was repeated three times and the pH and color value of LE

80 were as follows: pH =  $4.38 \pm 0.01$ , CIE L\* value =  $47.90 \pm 0.04$ , CIE a\* value =  $-4.86 \pm 0.02$ ,  
81 and CIE b\* value =  $35.37 \pm 0.03$ .

82

### 83 *Total phenolic content*

84 The amount of total phenols in LE and AA was determined using the Folin-Ciocalteu  
85 spectrophotometric method (Singleton & Rossi, 1965). The LE and AA solutions were  
86 prepared at a concentration of 5 mg/mL in 80% ethanol and DW, respectively. Gallic acid was  
87 used as a standard, and the total phenolic content of LE and AA was expressed as gallic acid  
88 equivalents (GAE) ( $\mu\text{g GAE/mL}$ ).

89

### 90 *1,1-Diphenyl-2-picrylhydrazyl (DPPH) radical scavenging activity*

91 DPPH radical scavenging activity of LE and AA was determined based on the method  
92 described by Jung and Sim (2019). The activity was determined using linear regression of the  
93 concentration-response curve of the percentage of DPPH radical inhibition versus various  
94 sample concentrations. The IC<sub>50</sub> value was expressed as the quantity of sample necessary to  
95 decrease the DPPH radical inhibition by 50%.

96

### 97 *2,2'-Azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS) radical scavenging activity*

98 The ABTS radical scavenging capacity of LE and AA was determined according the  
99 method described by Omoba et al. (2015). The activity was determined using linear regression  
100 of the concentration-response curve of the percentage of ABTS radical inhibition versus  
101 various sample concentrations. The IC<sub>50</sub> value was expressed as the quantity of sample  
102 necessary to decrease the ABTS radical inhibition by 50%.

103

### 104 *Metal chelating activity*

105 The chelation of ferrous ions by LE and AA was estimated according to the method described  
106 by Decker and Welch (1990). The activity was determined using linear regression of the  
107 concentration-response curve of the percentage of ferrous ion chelation versus various sample  
108 concentrations. The IC<sub>50</sub> value was expressed as the quantity of sample required to chelate Fe<sup>2+</sup>  
109 ions by 50%.

110

#### 111 *Preparation of semi-dried restructured jerky*

112 Fresh pork ham (*Musculus semimembranosus*, *Musculus semitendinosus*, *Musculus biceps*  
113 *femoris*) was obtained from a local market (Jeonju, Korea) and ground (Φ-8 mm). Binding  
114 meat batter was prepared by mixing the ground meat (20%) with a gelatin solution (0.2%  
115 phosphate and 1% duck skin gelatin) for 1 min in a silent cutter (Nr-963009, Hermann Scharfen  
116 GmbH & Co., Postfach, Germany). Semi-dried restructured jerky batter was obtained by  
117 mixing the binding meat batter and ground meat (80%) with ice water (10%), soy source (3%),  
118 sugar (2.0%), salt (1.2%), carrageenan (0.3%), black pepper powder (0.15%), garlic powder  
119 (0.15%), onion powder (0.15%), and sodium nitrite (0.005%). The amount of each ingredient  
120 was calculated relative to the total ground meat weight. The binding meat batter, ground meat,  
121 and other ingredients were mixed in a silent cutter for 2 min. The resulting jerky batter was  
122 divided into six batches. Each batch of samples comprised of restructured jerkies with different  
123 levels of loquat leaf extract powder (0, 0.15, and 0.3%) with or without ascorbic acid (0 and  
124 0.05%). The homogenized meat jerky batter was stuffed into a cellulose casing (Viskase Sale,  
125 Chicago, IL, USA: Φ- 20 mm) and each jerky was prepared as 20-cm-long pieces. Samples  
126 dried at 55 °C for 90 min in a chamber (MAXi3501 chamber, Kerres, Postfach, Germany) were  
127 removed from the casing, and the drying process was carried out as follows: 55 °C (30 min) →  
128 65 °C (180 min) → 80 °C (60 min) (Kim et al., 2020). The semi-dried restructured jerkies were  
129 vacuum-packed in polyethylene bags using a vacuum packager (HFV-500, Fufee Inc,

130 Hwaseong, Korea) and placed at room temperature for 180 days. The samples were taken on  
131 1, 20, 45, 90, 135 and 180<sup>th</sup> day for different quality parameters. This study was independently  
132 repeated thrice (three batches).

133

#### 134 *Sulfhydryl concentration*

135 The sulfhydryl concentration was measured by estimating protein oxidation using the  
136 method described by Berardo et al. (2015). The sulfhydryl group was detected via its reaction  
137 with 5,5'-dithio-bis-2-nitrobenzoic acid to form 5-mercapto-2-nitrobenzoic acid, and the  
138 absorbance was measured at 412 nm using a microplate reader (SpectraMax Plus 384,  
139 Molecular Devices Inc., CN, USA). Molar extinction coefficient of 14,000 mol/(L·cm) was  
140 used to calculate the sulfhydryl concentration in the jerkies, which was expressed as nmol/mg  
141 protein concentration.

142

#### 143 *Thiobarbituric acid-reactive substances (TBARS)*

144 To analyze lipid oxidation, the TBARS value was measured according to the method of  
145 Tarladgis et al. (1960). Each restructured jerky sample (5 g) was homogenized with 50 mL of  
146 distilled water (DW) and 0.2 mL of 0.3% butylated hydroxytoluene (Sigma-Aldrich, St.  
147 Louis, MO, USA) in methanol (Daejung Co., Goryeong, Korea), and was transferred to a  
148 distillation flask. The homogenate was distilled with 47.5 mL of DW, 2.5 mL of 4 N HCl  
149 (Samchun Co., Seoul, Korea), and 1 mL of anti-foaming agent (KMK073, Shin-Etsu Silicone  
150 Co., Ltd., Seoul, Korea), and 30 mL of the distillate was collected therefrom. Next, 5 mL of  
151 0.02 M TBA (Sigma-Aldrich) in 90% acetic acid (Junsei Chemical Co., Ltd., Tokyo, Japan)  
152 was added to each test tube containing 5 mL of the distillate and mixed. The tubes were capped  
153 and heated for 35 min at 100 °C and subsequently cooled under tap water (15 °C). Absorbance  
154 of the supernatant was measured at 538 nm using a microplate reader (SpectraMax Plus 384,



155 Molecular Devices Inc.). The amount of malondialdehyde (MDA) was calculated using a  
156 standard curve of 1,1,3,3-tetraethoxypropane, and the TBARS value was reported as mg  
157 MDA per kg of sample.

158

#### 159 *pH*

160 Approximately 5 g of the sample was homogenized for 1 min in 20 mL DW; pH of the  
161 homogenate was determined using a pH meter (Mettler-Toledo GmbH, Schwerzenbach,  
162 Switzerland) calibrated with pH at 4.7, and 10 standard solutions (Mettler-Toledo GmbH).

163

#### 164 *Color*

165 The color of restructured jerkies during storage was measured as CIE (International  
166 Commission on Illumination) L\*a\*b\* values by the colorimeter (CR-410, Minolta, Tokyo,  
167 Japan, D<sub>65</sub> light source, 2° observer). The instrument was calibrated using a standard white  
168 plate (L\* = + 97.83, a\* = - 0.43, and b\* = + 1.98).

169

#### 170 *Shear force*

171 The sample was cut into 3 cm lengths dimension for shear force analysis, and shear force  
172 was determined using a texture analyzer (TA-XT plus, Stable Micro Systems, Surrey, UK),  
173 and technical replicates were nine per sample.

174

#### 175 *Moisture content and water activity*

176 The moisture content of semi-dried restructured jerkies was measured using the AOAC  
177 (2000) method. The jerky was ground for the measurement of water activity, which was  
178 determined using a water activity meter (Novasina, Labmaster-aw, Lachen, Switzerland).

179

#### 180 *Microbial analysis*

181 On days 1, 20, 45, 90, 135, and 180 of storage, 6 g of jerky sample was aseptically placed  
182 into a sterile stomacher bag with 225 mL of 0.1% peptone water and then homogenized for 3  
183 min using the Stomacher Bag Mixer® 400 (Interscience Co., France). These homogenates were  
184 serially diluted with sterile saline solution. For microbial analysis, 1 mL of the diluted sample  
185 was inoculated on 3M Petrifilm plates (3M Microbiology, Saint Paul, MN, USA) to determine  
186 aerobic plate counts, coliform/*Escherichia coli* count, and yeast and mold count. For total  
187 aerobic bacteria and coliform/*E. coli*, plates were incubated at 37 °C for 48 h, and for yeast and  
188 mold, they were incubated at 25 °C for 5 days in an incubation chamber (J070217, Jeio Tech,  
189 Korea). Each microbial count was recorded as log colony-forming units per gram (log CFU/g).

190

#### 191 *Statistical analysis*

192 A t-test ( $p < 0.05$ ) was performed to compare total phenolic content and antioxidant  
193 activities between LE and AA, and leaf extract and ascorbic acid were considered as fixed  
194 terms and random terms were replicates. For the determination of jerky quality, LE  
195 concentration, addition of AA, and storage periods were considered as fixed terms and  
196 replicates were considered as random terms. A multifactorial analysis of variance, using the  
197 general linear model (GLM), was applied to determine the effects of LE concentration (0, 0.15,  
198 and 0.3%), with or without AA (0 and 0.05%), on storage period (1, 20, 45, 90, 135, and 180  
199 days). The data were analyzed using a two-way analysis of variance with treatments and

200 storage, using Duncan's multiple range test ( $p<0.05$ ) with the SPSS statistical software  
201 program (SPSS Ver. 20.0, IBM, Chicago, IL, USA).

202

## 203 **Results and Discussion**

204 *Total phenolic content and antioxidant activity of loquat leaf extract powder and ascorbic acid*

205 Table 1 shows the total phenolic content and antioxidant activities (DPPH radical  
206 scavenging, ABTS radical scavenging, and metal chelating activities) of LE and AA. The total  
207 phenolic content of LE was 533.38  $\mu\text{g GAE/mL}$ , whereas AA had no total phenolic content.  
208 The half maximal inhibitory concentration ( $\text{IC}_{50}$ ) of metal chelating activity of LE was 5.71  
209  $\text{mg/mL}$ , whereas AA had no metal chelating activity. The  $\text{IC}_{50}$  of DPPH radical scavenging  
210 activity of LE was lower than that of AA ( $p<0.05$ ). Moreover, the  $\text{IC}_{50}$  of ABTS radical  
211 scavenging activity of AA was lower than that of LE ( $p<0.05$ ). Various analyses were used to  
212 evaluate the antioxidant activity of LE and AA, and the results showed the antioxidant activities  
213 to be comparable to that reported in previous studies (Kim et al., 2019; Lee & Kim, 2009).  
214 Phenolic compounds are secondary metabolites of plants and are associated with antioxidant  
215 activities (Hwang et al., 2010). Generally, the antioxidant activity of phenolic compounds is  
216 based on their metal ion chelation and hydrogen donating abilities (Wijekoon et al., 2011).  
217 Phenolic hydroxyl groups bind heavy metals that cause free radical formation, thereby  
218 inhibiting lipid peroxidation (Kısa et al., 2016). AA is a strong antioxidant, owing to its  
219 preventive effect on oxidation of other compounds following the donation of electrons (Kim et  
220 al., 2013). In the present study, AA showed an excellent radical scavenging activity but no  
221 metal chelating activity. Since LE has metal chelating effect, a combination of LE and AA may  
222 be expected to have a synergistic effect as an oxidation inhibitor in semi-dried restructured  
223 jerky during storage.

224

225 *Sulfhydryl concentration and TBARS activity*

226 The sulfhydryl concentration and TBARS values of semi-dried restructured jerky prepared  
227 with LE and AA are shown in Table 2. When protein oxidation occurs, amino acid structures  
228 undergo some changes, such as formation of disulfide bridges, consequently leading to  
229 decreased sulfhydryl concentration (Turgut et al., 2017). The sulfhydryl concentration of jerky  
230 with LE and AA were higher ( $p<0.05$ ) than that of the control at 1 day of storage. The  
231 sulfhydryl concentration of jerky samples decreased during storage ( $p<0.001$ ) and significantly  
232 was high concentration related with an increase in LE concentration and the addition of AA  
233 ( $p<0.05$ ). Moreover, the combination (LE  $\times$  AA) was found to be significantly effective for  
234 sulfhydryl concentration in jerky during storage ( $p<0.001$ ); the highest sulfhydryl  
235 concentration was found in LE 0.3-AA, throughout the storage period ( $p<0.05$ ). The inhibited  
236 reduction of sulfhydryl concentration could be explained by the phenolic compounds of LE  
237 and antioxidant activities of LE and AA (Table 1). In general, hydroxyl groups of phenolic  
238 compounds bind to proteins, forming complexes, which in turn inhibit protein oxidation  
239 (Hoffman et al., 2014). Similar results to this study were reported by Fourati et al. (2020).  
240 Minced beef without any addition showed the lowest sulfhydryl concentration when compared  
241 with the sample added with antioxidants at 0 storage day. It is difficult to explain exactly why  
242 there was a difference in sulfhydryl concentration from day 0 or 1 of storage. However, if  
243 oxidation during the manufacturing process of meat products was inhibited by antioxidants,  
244 the oxidation values (sulfhydryl concentration or TBARS value) could be different at the beginning  
245 of storage. Of course, additional research is needed for more detailed explanation. According  
246 to Fourati et al. (2020), sulfhydryl concentration of minced beef was also significantly  
247 decreased during storage. Xu et al. (2018) reported that the addition of mulberry polyphenol

248 inhibited a decrease of sulfhydryl concentration in dried minced pork slices during the drying  
249 process and explained that phenolic compounds weaken protein oxidation in meat products.  
250 Moreover, metal chelation or DPPH and ABTS radical scavenging activity of LE and AA  
251 against free radicals was related to high sulfhydryl concentration of treatment compared to that  
252 in control.

253 The MDA level, representing the formation of secondary lipid oxidation products and  
254 measured as TBARS, indicate the progression of lipid oxidation in meat and meat products  
255 (Turgut et al., 2017). On 1 day of storage, all treatments added LE and AA, either alone or in  
256 combination, possessed lower TBARS value than control ( $p<0.05$ ). The TBARS values of all  
257 jerkies were increased during the storage period ( $p<0.05$ , Table 2). The jerky prepared with LE  
258 and AA, either alone or in combination, had lower TBARS values than the control during the  
259 storage period, thus indicating that LE, AA, and the combination (LE  $\times$  AA) had a significant  
260 effect on TBARS values ( $p<0.001$ ). Lipid oxidation can occur via three mechanisms: (i) free-  
261 radical chain reaction, (ii) metal- or enzyme- catalyzed oxidation, and (iii) photo-oxidation  
262 (Domínguez et al., 2019). The free-radical chain reaction is the most important mechanism of  
263 lipid oxidation in meat or meat products (Amaral et al., 2018). The iron present in myoglobin  
264 is a potent catalyst in various stages of lipid oxidation (Amaral et al., 2018; Domínguez et al.,  
265 2019). In the present study, as LE showed a strong metal chelating activity, its chelating activity  
266 with iron ions could possibly help retard lipid oxidation (Kong et al., 2010). Lee et al. (2011)  
267 had shown that increasing kimchi ethanolic extract concentration retarded the TBARS value of  
268 refrigerated cooked pork over 14 days of storage due to the strong metal-chelating activity. Our  
269 results indicated that the addition of LE and AA inhibited protein and lipid oxidation in semi-  
270 dried restructured jerky during storage, resulting in LE and AA being used as a radical

271 scavenger and chelating agent in foodstuff. However, further study is needed to maximize the  
272 usability of LE to retard lipid oxidation value.

273

#### 274 *pH and color*

275 The pH and color of the semi-dried restructured jerky, formulated with LE and AA, during  
276 storage are shown in Table 3. All treatments and control showed significantly decreased pH  
277 values during storage ( $p < 0.05$ ). The pH values of semi-dried restructured jerky significantly  
278 ( $p < 0.05$ ) decreased with an increase in LE concentration and addition of AA on day 1, and the  
279 combination (LE  $\times$  AA) had a significant effect on the pH values. When we measured the pH  
280 of LE and AA in DW, the value was 4.38 and 2.02, respectively (data not shown). The pH of  
281 LE and AA might have been affected by the initial pH of the semi-dried restructured jerky.  
282 According to Bower et al., (2003), the pH of vacuum-packaged meat products decreases with  
283 storage.

284 As shown in Table 3, the control and treatments had significantly ( $p < 0.05$ ) increased CIE  
285  $b^*$  and decreased CIE  $a^*$  values during storage, except for LE 0.3-AA ( $p > 0.05$ ). During the  
286 initial days of storage, there was no significant difference in CIE  $a^*$  value between LE 0.15-  
287 AA and control ( $p > 0.05$ ). At the end of the storage period, LE 0.15-AA had a higher CIE  $a^*$   
288 value than the control ( $p < 0.05$ ). The jerkies prepared with AA showed a higher CIE  $a^*$  value  
289 compared to those prepared with same amount of LE during storage ( $p < 0.05$ ), indicating that  
290 the addition of AA has a significant effect on the CIE  $a^*$  values ( $p < 0.001$ ). Lipid oxidation may  
291 catalyze pigment oxidation, and free radicals produced during oxidation may denature the  
292 myoglobin or oxidize the iron atoms, leading to a negative change in the color of products  
293 (Selani et al., 2011). The high antioxidant activity of LE is related to delayed or prevented  
294 oxidative reactions (Liu et al., 2016). AA generally improves color stability and leads to

295 declined myoglobin oxidation in meat products (Hwang et al., 2013). Since jerkies prepared  
296 with LE and AA had low TBARS values compared to the control during storage (Table 2), the  
297 trend of inhibitory change of CIE a\* values may be due to inhibition of lipid oxidation.  
298 Additionally, the change of color values in the initial days of storage might be affected by the  
299 color of LE (CIE L\*-value = 47.90, CIE a\*-value = -4.86, and CIE b\*-value = 35.37; data not  
300 shown).

301

### 302 *Moisture content, water activity, and shear force*

303 Moisture content and water activity in semi-dried restructured jerky, prepared with LE and  
304 AA, during storage are shown in Table 4. Both the parameters varied from 44.71 to 45.47%  
305 and from 0.882 to 0.890, respectively, on day 1, which matched the normal moisture content  
306 and water activity range of 20–50% and 0.82–0.91, respectively, of semi-dried jerky (Chen et  
307 al., 2002). The moisture content of jerkies significantly decreased over storage days ( $p < 0.001$ ),  
308 except for samples from LE 0.3-AA (the reduction was not significant,  $p > 0.05$ ). At the end of  
309 the storage period, LE 0.3-AA, LE 0.15-AA, and AA showed higher ( $p < 0.05$ ) moisture content,  
310 compared to the same amount of LE, indicating that AA had a significant effect on moisture  
311 content ( $p < 0.01$ ). Similarly, the water activity of jerky samples significantly decreased during  
312 storage ( $p < 0.001$ ; Table 4). However, the increase in LE concentration had a significant effect  
313 on the inhibited water activity in the jerky during the storage period ( $p < 0.001$ ). Similar results  
314 were reported by Kim et al. (2012), who showed that the moisture content and water activity  
315 of pork jerky decreased during storage, leading to a tough texture of jerky during distribution  
316 and storage. Additionally, Choi et al. (2007) had reported that water is diffused and  
317 dehumidified from the products in packaging; therefore, moisture content and water activity of

318 products may decrease during storage. Jerky requires a stable moisture content and water  
319 activity to avoid changes in quality properties during storage (Lim et al., 2013).

320 The shear force values of semi-dried restructured jerky increased with increasing LE  
321 concentration ( $p<0.001$ , Table 4). The shear force value in the control and treatment groups  
322 significantly ( $p<0.05$ ) increased during storage, and the sharpest incline occurred in the control.  
323 At the end of the storage period, LE 0.15-AA, LE 0.15, and AA had lower shear force than the  
324 control ( $p<0.05$ ). The increased shear force could be explained by increased protein oxidation  
325 and moisture content during storage. Disulfide bond formation caused by protein oxidation  
326 might weaken protein solubility, resulting in protein aggregation and complex formation (Xu  
327 et al., 2018). In this study, the addition of LE and AA inhibited protein and lipid oxidation;  
328 thus these might have also inhibited the increase in shear force in semi-dried restructured jerky  
329 during storage. Additionally, Li et al. (2014) had reported that the moisture content was  
330 extremely significantly correlated with shear force, and shear force value can represent the  
331 parameters of both moisture content and texture quality. The high moisture content of LE 0.15-  
332 AA, LE 0.15, and AA compared to that in the control seemed to be related to their low shear  
333 force at the end of the storage period (Table 4). All the results together suggested that a  
334 combination of LE and AA could minimize the texture deterioration of semi-dried restructured  
335 jerky caused by protein and lipid oxidation.

336

### 337 *Microbial analysis*

338 Results obtained from the microbial analyses of semi-dried restructured jerky prepared with  
339 LE and AA during storage are shown in Table 5. The total aerobic bacteria count of all jerkies  
340 after storage of 45 days was detected, and the total aerobic bacteria count of all jerkies increased  
341 during storage, regardless of control and treatments ( $p<0.05$ ). Coliform/*E. coli*, yeast, and



342 molds were not detected in all jerkies during storage (date not shown). At present, the  
343 established limit for total aerobic bacteria counts in protective dried meat, including jerky, is  
344 below 5.0 log CFU/g (Gaikwad et al., 2020); in this study, after a 135-day storage period, the  
345 jerky was still in agreement with these criteria (Table 5). Kim et al. (2013) had reported similar  
346 results of increased total aerobic bacterial count, in addition to no detected mold and coliform  
347 bacterial populations in pork jerky during a 2-month storage period, regardless of the control  
348 (without leek extract); the jerky was prepared with leek extract. Lim et al. (2012) had also  
349 reported that the total plate count of sun-dried beef jerky increased regardless of the addition  
350 of *Citrus junos seib* and *Prunus mume* extracts during storage. Several studies have noted that  
351 the fermented loquat leaf ethanol extract possesses antimicrobial activity and that the  
352 differences in the quantity of loquat leaf extract can result in differences in their antimicrobial  
353 activity, which can consequently affect microbial growth on meat products with loquat leaf  
354 extract (Dhiman et al., 2021; Fu et al., 2019; Liu et al., 2016). Further studies would be required  
355 to test the changes in microbial growth in semi-dry structured jerky supplemented with various  
356 concentrations of LE during storage.

357

## 358 **Conclusions**

359 This study reported that LE and AA have high antioxidant activities, as shown by their metal  
360 chelating, as well as DPPH and ABTS radical scavenging activities. The combination of LE  
361 and AA retarded protein and lipid oxidation (as per sulfhydryl concentration and TBARS  
362 value) in semi-dried restructured jerky, during storage. The combination of 0.15% loquat leaf  
363 extract powder and 0.05% ascorbic acid was determined to be applicable as a natural  
364 antioxidant for preserving the quality of semi-dried restructured jerky, and it has a potential  
365 application in preserving meat and meat products.

366

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489

## Table legend

490

491

492 Table 1. Total phenolic content and antioxidant activities of loquat leaf extract (LE) and ascorbic acid  
493 (AA)

494 Table 2. Protein oxidation and lipid oxidation in semi-dried restructured jerky prepared with loquat  
495 leaf extract (LE) and ascorbic acid (AA) during storage

496 Table 3. pH and color in semi-dried restructured jerky prepared with loquat leaf extract (LE) and  
497 ascorbic acid (AA) during storage

498 Table 4. Moisture content, water activity, and shear force in semi-dried restructured jerky prepared  
499 with loquat leaf extract (LE) and ascorbic acid (AA) during storage

500 Table 5. Microbial analysis in semi-dried restructured jerky prepared with loquat leaf extract (LE) and  
501 ascorbic acid (AA) during storage

502

503

504 Table 1. Total phenolic content and antioxidant activities of loquat leaf extract (LE) and ascorbic acid (AA)

Treatment	Loquat leaf extract	Ascorbic acid
Total phenolic content ( $\mu\text{g GAE}^1/\text{mL}$ )	$533.38 \pm 5.49$	NE <sup>2)</sup>
DPPH radical scavenging activity ( $\text{IC}_{50}$ value mg/mL)	$0.06 \pm 0.00^b$	$0.10 \pm 0.00^a$
ABTS radical scavenging activity ( $\text{IC}_{50}$ value mg/mL)	$2.87 \pm 0.04^a$	$0.61 \pm 0.01^b$
Metal chelating activity ( $\text{IC}_{50}$ value mg/mL)	$5.71 \pm 0.03$	NE

505 All values are mean  $\pm$  standard error of three replicates.

506 <sup>a-b</sup> Means within a row with different letters are significantly different.

507 <sup>1)</sup>GAE: gallic acid equivalents

508 <sup>2)</sup>NE: values were not estimated.

509



510 Table 2. Protein oxidation and lipid oxidation values in semi-dried restructured jerky prepared with loquat leaf extract (LE) and ascorbic acid (AA)  
 511 during storage

Treatment <sup>1)</sup>	Storage (days)					
	1	20	45	90	135	180
<b>Sulfhydryl concentration (nmol/mg protein)</b>						
Control	48.27 ± 0.13 <sup>Fa</sup>	38.61 ± 0.15 <sup>Fb</sup>	33.80 ± 0.09 <sup>Fc</sup>	28.99 ± 0.06 <sup>Fd</sup>	24.33 ± 0.04 <sup>Fe</sup>	18.91 ± 0.06 <sup>Ff</sup>
AA	51.82 ± 0.29 <sup>Da</sup>	45.76 ± 0.31 <sup>Eb</sup>	43.84 ± 0.06 <sup>Dc</sup>	38.88 ± 0.13 <sup>Dd</sup>	32.52 ± 0.21 <sup>Ee</sup>	26.14 ± 0.09 <sup>Ef</sup>
LE0.15	51.23 ± 0.07 <sup>Ea</sup>	46.76 ± 0.35 <sup>Db</sup>	39.41 ± 0.12 <sup>Ec</sup>	35.81 ± 0.15 <sup>Ed</sup>	33.08 ± 0.09 <sup>De</sup>	29.37 ± 0.04 <sup>Df</sup>
LE0.15-AA	55.80 ± 0.33 <sup>Ca</sup>	51.49 ± 0.24 <sup>Cb</sup>	44.48 ± 0.06 <sup>Cc</sup>	39.82 ± 0.13 <sup>Cd</sup>	35.67 ± 0.14 <sup>Ce</sup>	32.08 ± 0.02 <sup>Cf</sup>
LE0.3	56.79 ± 0.03 <sup>Ba</sup>	53.55 ± 0.25 <sup>Bb</sup>	46.73 ± 0.20 <sup>Bc</sup>	44.48 ± 0.06 <sup>Bd</sup>	39.73 ± 0.10 <sup>Be</sup>	35.67 ± 0.14 <sup>Bf</sup>
LE0.3-AA	62.77 ± 0.05 <sup>Aa</sup>	60.45 ± 0.30 <sup>Ab</sup>	56.08 ± 0.38 <sup>Ac</sup>	53.68 ± 0.14 <sup>Ad</sup>	48.06 ± 0.12 <sup>Ae</sup>	44.51 ± 0.34 <sup>Af</sup>
<b>TBARS (mg MDA/kg)</b>						
Control	0.36 ± 0.00 <sup>Ac</sup>	0.39 ± 0.01 <sup>Ad</sup>	0.41 ± 0.01 <sup>Ac</sup>	0.42 ± 0.01 <sup>Ab</sup>	0.42 ± 0.01 <sup>Ab</sup>	0.45 ± 0.01 <sup>Aa</sup>
AA	0.35 ± 0.00 <sup>Ce</sup>	0.36 ± 0.00 <sup>Bcd</sup>	0.40 ± 0.00 <sup>BCc</sup>	0.41 ± 0.01 <sup>Bb</sup>	0.40 ± 0.00 <sup>CDb</sup>	0.44 ± 0.01 <sup>Ba</sup>
LE0.15	0.35 ± 0.00 <sup>Be</sup>	0.36 ± 0.00 <sup>Bd</sup>	0.40 ± 0.00 <sup>Bc</sup>	0.40 ± 0.01 <sup>BCc</sup>	0.41 ± 0.00 <sup>Bb</sup>	0.44 ± 0.01 <sup>BCa</sup>
LE0.15-AA	0.35 ± 0.00 <sup>Ce</sup>	0.36 ± 0.00 <sup>BCd</sup>	0.39 ± 0.01 <sup>Dc</sup>	0.40 ± 0.01 <sup>CDbc</sup>	0.40 ± 0.00 <sup>Db</sup>	0.43 ± 0.01 <sup>DEa</sup>

LE 0.3	0.35 ± 0.00 <sup>Ce</sup>	0.36 ± 0.00 <sup>Cd</sup>	0.39 ± 0.00 <sup>CDc</sup>	0.40 ± 0.01 <sup>DEc</sup>	0.41 ± 0.01 <sup>BCb</sup>	0.43 ± 0.01 <sup>CDa</sup>
LE 0.3-AA	0.35 ± 0.00 <sup>Cd</sup>	0.35 ± 0.00 <sup>Dd</sup>	0.39 ± 0.00 <sup>Dc</sup>	0.39 ± 0.01 <sup>Ec</sup>	0.40 ± 0.01 <sup>Db</sup>	0.43 ± 0.01 <sup>Ea</sup>

	Storage	LE	AA	Storage × LE	Storage × AA	LE × AA	Storage × LE × AA
<b>Sulphydryl concentration</b>							
<i>p</i> -value	***	***	***	***	***	***	***
<b>TBARS</b>							
<i>p</i> -value	***	***	***	***	**	***	**

512 All values are mean ± standard error of three replicates.

513 <sup>A-F</sup> Means sharing different letters in the same column are significantly different.

514 <sup>a-f</sup> Means sharing different letters in the same row are significantly different.

515 \*\* *p* < 0.01, \*\*\* *p* < 0.001.

516 <sup>1)</sup> Control, no antioxidant; AA, added 0.05% ascorbic acid; LE 0.15, added 0.15 loquat leaf extract; LE 0.15-AA, added 0.15% loquat leaf extract and

517 0.05% ascorbic acid; LE 0.3, added 0.3% loquat leaf extract; LE0.3-AA, added 0.3% loquat leaf extract and 0.05% ascorbic acid.

518

519 Table 3. pH and color in semi-dried restructured jerky prepared with loquat leaf extract (LE) and ascorbic acid (AA) during storage

Treatment <sup>1)</sup>	Storage (days)					
	1	20	45	90	135	180
<b>pH</b>						
Control	6.15 ± 0.01 <sup>AA</sup>	6.12 ± 0.01 <sup>Ab</sup>	6.12 ± 0.01 <sup>Ab</sup>	6.11 ± 0.01 <sup>Abc</sup>	6.11 ± 0.01 <sup>Ac</sup>	6.10 ± 0.01 <sup>Ac</sup>
AA	6.06 ± 0.01 <sup>Da</sup>	6.06 ± 0.01 <sup>Ca</sup>	6.06 ± 0.01 <sup>Ba</sup>	6.06 ± 0.01 <sup>Ba</sup>	6.06 ± 0.01 <sup>Ba</sup>	6.01 ± 0.01 <sup>Db</sup>
LE0.15	6.09 ± 0.01 <sup>Ba</sup>	6.07 ± 0.01 <sup>BCb</sup>	6.07 ± 0.01 <sup>Bb</sup>	6.06 ± 0.01 <sup>Bc</sup>	6.06 ± 0.01 <sup>Bc</sup>	6.05 ± 0.01 <sup>Bc</sup>
LE0.15-AA	6.05 ± 0.01 <sup>Ea</sup>	6.04 ± 0.01 <sup>Db</sup>	6.04 ± 0.01 <sup>Cb</sup>	6.02 ± 0.01 <sup>Cc</sup>	6.00 ± 0.01 <sup>Dd</sup>	6.00 ± 0.01 <sup>Ee</sup>
LE0.3	6.08 ± 0.01 <sup>Ca</sup>	6.08 ± 0.01 <sup>Ba</sup>	6.06 ± 0.01 <sup>Bb</sup>	6.06 ± 0.01 <sup>Bb</sup>	6.05 ± 0.01 <sup>Bc</sup>	6.03 ± 0.01 <sup>Cd</sup>
LE0.3-AA	6.04 ± 0.01 <sup>Fa</sup>	6.03 ± 0.01 <sup>Db</sup>	6.02 ± 0.01 <sup>Dc</sup>	6.02 ± 0.01 <sup>Cc</sup>	6.02 ± 0.01 <sup>Cc</sup>	5.99 ± 0.01 <sup>Ed</sup>
<b>CIEL*</b>						
Control	48.72 ± 0.38 <sup>ABc</sup>	50.31 ± 0.19 <sup>Ad</sup>	50.65 ± 0.18 <sup>Ad</sup>	51.57 ± 0.18 <sup>Ac</sup>	53.22 ± 0.21 <sup>Ab</sup>	54.18 ± 0.17 <sup>Aa</sup>
AA	49.07 ± 0.19 <sup>Ac</sup>	49.13 ± 0.19 <sup>Bc</sup>	50.38 ± 0.08 <sup>Ab</sup>	50.58 ± 0.30 <sup>Bb</sup>	50.93 ± 0.13 <sup>Bb</sup>	51.68 ± 0.38 <sup>Ba</sup>
LE0.15	47.67 ± 0.25 <sup>CDd</sup>	48.03 ± 0.22 <sup>Cd</sup>	49.33 ± 0.11 <sup>Cc</sup>	49.33 ± 0.15 <sup>Cc</sup>	50.10 ± 0.09 <sup>Cb</sup>	51.68 ± 0.38 <sup>Ba</sup>
LE0.15-AA	48.14 ± 0.08 <sup>BCc</sup>	48.23 ± 0.09 <sup>Cc</sup>	49.29 ± 0.14 <sup>Bb</sup>	49.66 ± 0.31 <sup>Cb</sup>	50.34 ± 0.18 <sup>Ca</sup>	50.46 ± 0.30 <sup>Ca</sup>
LE0.3	46.43 ± 0.19 <sup>Ec</sup>	46.82 ± 0.13 <sup>Dc</sup>	46.84 ± 0.15 <sup>Dc</sup>	47.96 ± 0.20 <sup>Db</sup>	48.02 ± 0.24 <sup>Db</sup>	48.90 ± 0.16 <sup>Da</sup>

LE 0.3-AA	47.13 ± 0.25 <sup>DEb</sup>	47.33 ± 0.28 <sup>Db</sup>	47.56 ± 0.12 <sup>Eb</sup>	47.69 ± 0.22 <sup>Db</sup>	47.48 ± 0.09 <sup>Eb</sup>	48.30 ± 0.18 <sup>Da</sup>
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**CIE a\***

Control	14.63 ± 0.12 <sup>Ba</sup>	14.35 ± 0.18 <sup>AA</sup>	13.80 ± 0.07 <sup>Bb</sup>	12.92 ± 0.22 <sup>Bc</sup>	12.48 ± 0.15 <sup>Bc</sup>	11.65 ± 0.27 <sup>Cd</sup>
AA	16.62 ± 0.33 <sup>AA</sup>	16.20 ± 0.19 <sup>Bab</sup>	16.02 ± 0.19 <sup>Ab</sup>	15.46 ± 0.07 <sup>Ac</sup>	14.73 ± 0.11 <sup>Ad</sup>	14.38 ± 0.14 <sup>Ad</sup>
LE 0.15	11.91 ± 0.33 <sup>Ca</sup>	11.56 ± 0.32 <sup>Dab</sup>	11.31 ± 0.18 <sup>Dabc</sup>	11.10 ± 0.11 <sup>Cbc</sup>	10.74 ± 0.07 <sup>Cc</sup>	9.52 ± 0.05 <sup>Dd</sup>
LE 0.15-AA	14.25 ± 0.14 <sup>Ba</sup>	13.33 ± 0.25 <sup>Cb</sup>	13.33 ± 0.13 <sup>Cb</sup>	12.79 ± 0.01 <sup>Bc</sup>	12.75 ± 0.12 <sup>Bc</sup>	12.56 ± 0.17 <sup>Bc</sup>
LE 0.3	9.76 ± 0.05 <sup>Ea</sup>	9.73 ± 0.09 <sup>Fa</sup>	9.53 ± 0.11 <sup>Fa</sup>	9.03 ± 0.22 <sup>Eab</sup>	8.81 ± 0.43 <sup>Db</sup>	8.73 ± 0.27 <sup>Eb</sup>
LE 0.3-AA	10.55 ± 0.12 <sup>D</sup>	10.54 ± 0.04 <sup>E</sup>	10.43 ± 0.10 <sup>E</sup>	10.40 ± 0.10 <sup>D</sup>	10.21 ± 0.05 <sup>C</sup>	10.04 ± 0.38 <sup>D</sup>

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**CIE b\***

Control	12.09 ± 0.03 <sup>Fe</sup>	14.56 ± 0.24 <sup>Ed</sup>	15.28 ± 0.24 <sup>Ed</sup>	16.17 ± 0.32 <sup>Ec</sup>	17.36 ± 0.32 <sup>Db</sup>	18.41 ± 0.23 <sup>Da</sup>
AA	14.63 ± 0.36 <sup>Ec</sup>	14.83 ± 0.14 <sup>Ec</sup>	15.80 ± 0.25 <sup>Eb</sup>	15.95 ± 0.16 <sup>Eb</sup>	16.64 ± 0.29 <sup>Da</sup>	17.05 ± 0.12 <sup>Ea</sup>
LE 0.15	15.77 ± 0.43 <sup>Dd</sup>	16.32 ± 0.15 <sup>Dcd</sup>	16.99 ± 0.05 <sup>Dbc</sup>	17.27 ± 0.11 <sup>Db</sup>	18.66 ± 0.42 <sup>Ca</sup>	19.09 ± 0.09 <sup>Ca</sup>
LE 0.15-AA	17.57 ± 0.15 <sup>Cb</sup>	17.79 ± 0.19 <sup>Cb</sup>	18.00 ± 0.19 <sup>Cb</sup>	18.22 ± 0.23 <sup>Cb</sup>	19.31 ± 0.40 <sup>Ca</sup>	19.62 ± 0.14 <sup>Ca</sup>
LE 0.3	19.76 ± 0.17 <sup>Bc</sup>	19.90 ± 0.11 <sup>Bc</sup>	20.01 ± 0.07 <sup>Bc</sup>	20.30 ± 0.24 <sup>Bbc</sup>	20.84 ± 0.33 <sup>Bab</sup>	20.95 ± 0.16 <sup>Ba</sup>
LE 0.3-AA	21.83 ± 0.22 <sup>A</sup>	21.97 ± 0.18 <sup>A</sup>	22.10 ± 0.18 <sup>A</sup>	22.23 ± 0.36 <sup>A</sup>	22.63 ± 0.35 <sup>A</sup>	22.67 ± 0.30 <sup>A</sup>

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<i>p</i> -value	Storage	LE	AA	Storage × LE	Storage × AA	LE × AA	Storage × LE × AA
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pH	***	***	***	***	***	***	***
CIEL*	***	***	***	***	***	***	*
CIE a*	***	***	***	***	**	***	NS
CIE b*	***	***	***	***	***	***	***

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520 All values are mean  $\pm$  standard error of three replicates.

521 <sup>A-F</sup> Means sharing different letters in the same column are significantly different.

522 <sup>a-f</sup> Means sharing different letters in the same row are significantly different.

523 \*  $p < 0.05$ , \*\*  $p < 0.01$ , \*\*\*  $p < 0.001$ ; NS, no significance.

524 <sup>1)</sup> Control, no antioxidant; AA, added 0.05% ascorbic acid; LE 0.15, added 0.15 loquat leaf extract; LE 0.15-AA, added 0.15% loquat leaf extract and  
525 0.05% ascorbic acid; LE 0.3, added 0.3% loquat leaf extract; LE0.3-AA, added 0.3% loquat leaf extract and 0.05% ascorbic acid.

526

527 Table 4. Moisture content, water activity, and shear force in semi-dried restructured jerky prepared with loquat leaf extract (LE) and ascorbic acid  
 528 (AA) during storage

Treatment <sup>1)</sup>	Storage (days)					
	1	20	45	90	135	180
<b>Moisture content (%)</b>						
Control	45.47 ± 0.33 <sup>a</sup>	44.70 ± 0.19 <sup>ab</sup>	44.20 ± 0.40 <sup>bc</sup>	43.72 ± 0.24 <sup>Cc</sup>	43.67 ± 0.17 <sup>c</sup>	42.53 ± 0.20 <sup>Dd</sup>
AA	45.29 ± 0.16 <sup>a</sup>	44.97 ± 0.28 <sup>ab</sup>	44.97 ± 0.09 <sup>ab</sup>	44.54 ± 0.13 <sup>Abc</sup>	44.23 ± 0.12 <sup>c</sup>	43.70 ± 0.14 <sup>ABd</sup>
LE0.15	45.11 ± 0.16 <sup>a</sup>	44.93 ± 0.16 <sup>ab</sup>	44.81 ± 0.55 <sup>abc</sup>	44.21 ± 0.07 <sup>ABbc</sup>	44.06 ± 0.05 <sup>c</sup>	43.13 ± 0.02 <sup>Cd</sup>
LE0.15-AA	44.81 ± 0.28 <sup>a</sup>	44.61 ± 0.14 <sup>ab</sup>	44.49 ± 0.12 <sup>ab</sup>	44.33 ± 0.10 <sup>Aabc</sup>	44.24 ± 0.11 <sup>bc</sup>	43.90 ± 0.17 <sup>Ac</sup>
LE0.3	44.99 ± 0.30 <sup>a</sup>	44.67 ± 0.15 <sup>a</sup>	44.57 ± 0.22 <sup>a</sup>	43.87 ± 0.09 <sup>BCb</sup>	43.55 ± 0.30 <sup>b</sup>	43.35 ± 0.20 <sup>BCb</sup>
LE0.3-AA	44.71 ± 0.05	44.63 ± 0.28	44.24 ± 0.47	44.09 ± 0.14 <sup>ABC</sup>	44.00 ± 0.26	44.16 ± 0.06 <sup>A</sup>
<b>Water activity</b>						
Control	0.890 ± 0.001 <sup>a</sup>	0.888 ± 0.003 <sup>Aa</sup>	0.886 ± 0.001 <sup>a</sup>	0.886 ± 0.002 <sup>Aa</sup>	0.869 ± 0.002 <sup>b</sup>	0.860 ± 0.001 <sup>Ec</sup>
AA	0.884 ± 0.001 <sup>a</sup>	0.887 ± 0.001 <sup>Aa</sup>	0.885 ± 0.001 <sup>a</sup>	0.874 ± 0.002 <sup>Bb</sup>	0.872 ± 0.001 <sup>b</sup>	0.862 ± 0.000 <sup>Dc</sup>
LE0.15	0.882 ± 0.003 <sup>a</sup>	0.880 ± 0.000 <sup>BCa</sup>	0.883 ± 0.002 <sup>a</sup>	0.874 ± 0.002 <sup>Bb</sup>	0.872 ± 0.000 <sup>b</sup>	0.864 ± 0.000 <sup>Cc</sup>
LE0.15-AA	0.882 ± 0.002 <sup>a</sup>	0.877 ± 0.001 <sup>Cab</sup>	0.877 ± 0.003 <sup>ab</sup>	0.872 ± 0.001 <sup>Bbc</sup>	0.869 ± 0.002 <sup>bc</sup>	0.866 ± 0.000 <sup>Bc</sup>

LE 0.3	0.886 ± 0.000 <sup>a</sup>	0.883 ± 0.001 <sup>ABa</sup>	0.885 ± 0.003 <sup>a</sup>	0.874 ± 0.003 <sup>Bb</sup>	0.874 ± 0.002 <sup>b</sup>	0.868 ± 0.000 <sup>Ab</sup>
LE 0.3-AA	0.884 ± 0.002 <sup>a</sup>	0.882 ± 0.006 <sup>ABa</sup>	0.884 ± 0.002 <sup>a</sup>	0.877 ± 0.002 <sup>Bab</sup>	0.870 ± 0.001 <sup>bc</sup>	0.868 ± 0.000 <sup>ABc</sup>

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**Shear force (N)**

Control	144.67 ± 0.36 <sup>De</sup>	150.49 ± 0.34 <sup>Dd</sup>	151.37 ± 0.26 <sup>Cd</sup>	154.31 ± 0.39 <sup>Ec</sup>	159.15 ± 0.23 <sup>Cb</sup>	165.30 ± 0.43 <sup>Ca</sup>
AA	145.33 ± 0.36 <sup>Df</sup>	147.14 ± 0.42 <sup>Ec</sup>	148.08 ± 0.29 <sup>Dd</sup>	151.01 ± 0.25 <sup>Fc</sup>	153.79 ± 0.22 <sup>Db</sup>	157.00 ± 0.31 <sup>Fa</sup>
LE 0.15	153.59 ± 0.09 <sup>Cf</sup>	154.47 ± 0.10 <sup>Ce</sup>	156.20 ± 0.01 <sup>Bd</sup>	158.62 ± 0.35 <sup>Cc</sup>	160.43 ± 0.46 <sup>Bb</sup>	161.73 ± 0.32 <sup>Da</sup>
LE 0.15-AA	154.97 ± 0.33 <sup>Bd</sup>	155.72 ± 0.29 <sup>Bcd</sup>	156.12 ± 0.16 <sup>Bc</sup>	157.14 ± 0.20 <sup>Db</sup>	158.80 ± 0.31 <sup>Ca</sup>	159.37 ± 0.41 <sup>Ea</sup>
LE 0.3	166.90 ± 0.27 <sup>Ae</sup>	167.04 ± 0.12 <sup>Ae</sup>	167.91 ± 0.18 <sup>Ad</sup>	169.18 ± 0.07 <sup>Bc</sup>	171.11 ± 0.31 <sup>Ab</sup>	172.60 ± 0.30 <sup>AA</sup>
LE 0.3-AA	167.52 ± 0.16 <sup>Ab</sup>	167.56 ± 0.24 <sup>Ab</sup>	168.23 ± 0.32 <sup>Ab</sup>	170.77 ± 0.46 <sup>AA</sup>	170.77 ± 0.41 <sup>AA</sup>	171.15 ± 0.44 <sup>Ba</sup>

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<i>p</i> -value	Storage	LE	AA	Storage × LE	Storage × AA	LE × AA	Storage × LE × AA
Moisture content	***	NS	**	NS	***	*	NS
Water activity	***	***	*	***	NS	NS	*
Shear force	***	***	***	***	***	***	***

529 All values are mean ± standard error of three replicates.

530 <sup>A-F</sup> Means sharing different letters in the same column are significantly different.

531 <sup>a-f</sup> Means sharing different letters in the same row are significantly different.

532 \*  $p < 0.05$ , \*\*  $p < 0.01$ , \*\*\*  $p < 0.001$ ; NS, no significance.

533 <sup>1)</sup> Control, no antioxidant; AA, added 0.05% ascorbic acid; LE 0.15, added 0.15 loquat leaf extract; LE 0.15-AA, added 0.15% loquat leaf extract and  
534 0.05% ascorbic acid; LE 0.3, added 0.3% loquat leaf extract; LE0.3-AA, added 0.3% loquat leaf extract and 0.05% ascorbic acid.

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536 Table 5. Microbial analysis in semi-dried restructured jerky prepared with loquat leaf extract (LE) and ascorbic acid (AA) during storage

Treatment <sup>1)</sup>	Storage (days)						
	1	20	45	90	135	180	
<b>Total aerobic bacteria (Log CFU/g)</b>							
Control	ND <sup>d2)</sup>	ND <sup>d</sup>	ND <sup>d</sup>	3.49 ± 0.06 <sup>c</sup>	4.07 ± 0.04 <sup>b</sup>	5.39 ± 0.06 <sup>a</sup>	
AA	ND <sup>d</sup>	ND <sup>d</sup>	ND <sup>d</sup>	3.47 ± 0.06 <sup>c</sup>	4.08 ± 0.04 <sup>b</sup>	5.40 ± 0.07 <sup>a</sup>	
LE0.15	ND <sup>d</sup>	ND <sup>d</sup>	ND <sup>d</sup>	3.52 ± 0.06 <sup>c</sup>	4.10 ± 0.04 <sup>b</sup>	5.38 ± 0.07 <sup>a</sup>	
LE0.15-AA	ND <sup>d</sup>	ND <sup>d</sup>	ND <sup>d</sup>	3.50 ± 0.05 <sup>c</sup>	4.07 ± 0.05 <sup>b</sup>	5.41 ± 0.07 <sup>a</sup>	
LE0.3	ND <sup>d</sup>	ND <sup>d</sup>	ND <sup>d</sup>	3.47 ± 0.05 <sup>c</sup>	4.09 ± 0.06 <sup>b</sup>	5.42 ± 0.07 <sup>a</sup>	
LE0.3-AA	ND <sup>d</sup>	ND <sup>d</sup>	ND <sup>d</sup>	3.54 ± 0.04 <sup>c</sup>	4.08 ± 0.03 <sup>b</sup>	5.41 ± 0.07 <sup>a</sup>	
<i>p</i> -value	Storage	LE	AA	Storage × LE	Storage × AA	LE × AA	Storage × LE × AA
Total aerobic bacteria ***		NS	NS	NS	NS	NS	NS

537 All values are mean ± standard error of three replicates.

538 <sup>a-d</sup> Means sharing different letters in the same row are significantly different.

539 \*\*\* *p* < 0.001; NS, no significance.

540 <sup>1)</sup>Control, no antioxidant; AA, added 0.05% ascorbic acid; LE 0.15, added 0.15 loquat leaf extract; LE 0.15-AA, added 0.15% loquat leaf extract and

541 0.05% ascorbic acid; LE 0.3, added 0.3% loquat leaf extract; LE0.3-AA, added 0.3% loquat leaf extract and 0.05% ascorbic acid.

542 <sup>2)</sup> ND, not detected with the detection limit of  $< 10^1$  CFU/g

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