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8 **Effects of Gochujang (Korean Red Pepper Paste) Marinade on Polycyclic Aromatic**
9 **Hydrocarbon Formation in Charcoal-grilled Pork Belly**

10

11 **Abstract**

12 Charcoal-grilling is a popular cooking method but causes the formation of polycyclic
13 aromatic hydrocarbons (PAHs), which can be harmful to human health. Gochujang marinade
14 is commonly used for flavoring meats during charcoal-grilling. However, the effects of this
15 marinade on PAHs formation during charcoal-grilling are unclear. Here, we evaluated the
16 effects of Gochujang marinade on the formation of 16 PAHs and inhibition rate of major
17 PAHs (benzo[a]anthracene, benzo[b]fluoranthene, and benzo[a]pyrene) in charcoal-grilled
18 pork belly. Pork belly without marinade (PBW) and marinated with Gochujang (PBG) were
19 stored for 10 days at 9°C under vacuum conditions and then charcoal-grilled to different
20 doneness (internal temperatures of 71°C and 81°C). Among 16 PAHs evaluated in this study,
21 14 PAHs were detected in charcoal-grilled pork belly, regardless of doneness. PAH formation
22 in charcoal-grilled pork belly was higher at an internal temperature of 81°C than at 71°C ($p <$
23 0.05). Initially, PBG showed reduced total PAH formation and lower percentages of three
24 major PAHs compared with PBW. Storage increased the inhibitory effects of PBG on the 16
25 PAHs, and the maximum reduction in total 16PAHs (63.06%) was observed with moderate
26 cooking (71°C) on day 10 ($p < 0.05$). Moreover, marinade and doneness showed a high
27 interaction with regard to PAH contents in charcoal-grilled pork belly ($p < 0.05$ - $p < 0.0001$).
28 Therefore, our findings suggested that marinating pork belly with Gochujang and grilling at
29 71°C could reduce the formation of 16 PAHs in charcoal-grilled pork belly.

30

31 **Keywords** polycyclic aromatic hydrocarbon, barbecue, natural antioxidant, storage,

32 doneness

33

34 **Introduction**

35 The use of barbecues or grilling has become increasingly popular in recent years in Korea.
36 High-temperature conditions and the use of charcoal during grilling substantially improve
37 flavor. However, undesirable substances, such as polycyclic aromatic hydrocarbons (PAHs)
38 and heterocyclic aromatic amines, are produced during charcoal-grilling (Wang et al., 2019).
39 PAHs are a large group of persistent organic compounds containing two or more fused
40 aromatic rings (Gong et al., 2018), which are considered potentially carcinogenic in humans
41 because of their genotoxic properties (Mejborn et al., 2019). Although the mechanisms
42 through which PAHs form in meats during grilling are not understood fully, some studies
43 have reported three mechanisms of PAH formation in grilled meats, as follows: (i) the
44 pyrolysis of organic compounds or fats in meat creates a layer on the meat containing PAHs;
45 (ii) the incomplete combustion of coals produces smoke, which attaches to the surface of
46 meat; and (iii) melting fats drip over the burning charcoal, causing the formation of PAHs
47 that return to the meat via smoke (García-Lomillo et al., 2017; Viegas et al., 2014; Wang et
48 al., 2019). Additionally, various factors affect the formation of PAHs in meat, including
49 cooking temperature, time, and fat content (Sahin et al., 2020). In particular, high doneness
50 and fat contents in meat can increase the PAH contents in grilled meats (Kim et al., 2021).
51 Based on concerns regarding the types of PAHs generated in grilled meat, the United States
52 Environmental Protection Agency listed 16 PAHs with carcinogenic and mutagenic effects,
53 namely, naphthalene, acenaphthene, acenaphthylene, fluorene, anthracene, phenanthrene,
54 fluoranthene, pyrene, benzo[a]anthracene, chrysene, benzo[b]fluoranthene,
55 benzo[k]fluoranthene, benzo[a]pyrene, dibenzo[a,h]anthracene, benzo[g,h,i]perylene, and

56 indeno[1,2,3-cd]pyrene (Gong et al., 2018). Although these PAHs are present in charcoal-
57 grilled meat at low levels, frequent consumption of charcoal-grilled meats over a long period
58 of time can be harmful to human health. Therefore, it is important to control the formation of
59 these 16 PAHs in grilled meats.

60 Various studies have evaluated methods to inhibit PAH formation in meats during cooking
61 (Farhadian et al., 2012; Gibis, 2007; Park et al., 2017). Cooking at lower temperatures and for
62 shorter times can inhibit PAH formation in meat (Farhadian et al., 2012). Additionally,
63 preventing melted fat from meats from dripping onto the heat source reduces benzo[a]pyrene
64 contents in grilled pork belly (Park et al., 2017). In addition, marinating with ingredients with
65 antioxidant properties can reduce PAHs formation during the cooking process (Gibis, 2007).
66 Marinating, a traditional cooking technique, is performed to improve the flavor and
67 tenderness of meat in order to satisfy consumer demand (Fasano et al., 2016). Marinating
68 meats with natural antioxidants, such as polyphenols and sulfhydryl compounds in beer, tea,
69 onion, garlic, and lemon, has been reported to reduce the concentrations of PAHs in meats
70 (Viegas et al., 2014; Wang et al., 2019).

71 Gochujang, Korean fermented red-pepper paste, is traditionally composed of red pepper,
72 grains (e.g., barley, rice, and/or wheat), and soybean Meju with water (Kwon et al., 2015).
73 Recent studies have reported that Gochujang exhibits bioactivities, including anti-
74 atherosclerotic, anti-obesity, and anticholesterol effects (Kim et al., 2019a; Shin et al., 2016;
75 Yang et al., 2018). These bioactivities are thought to be related to the various biological
76 compounds of Gochujang, including polyphenol compounds Meju and capsaicin derivatives
77 in red pepper (Reyes-Escogido et al. 2011; Yang et al., 2018). Koreans enjoy Gochujang as a
78 seasoning for charcoal-grilled pork belly and typically eat pork belly with dipping in
79 Gochujang or cook pork belly on the grill after marinating with Gochujang (Kwon et al.,

80 2015). According to Chung et al. (2009), Gochujang contained high phenolic acids
81 compared to other phenol contents, which showed greater PAH inhibitory effect than other
82 phenolic compounds, such as flavonoids (Wang et al., 2019). So, Gochujang can be used as
83 effective marinades for inhibiting PAH formation in grilled meat. However, the effects of
84 Gochujang marinade on PAHs formation during charcoal-grilling have not been reported.

85 Therefore, in this study, we evaluated the effects of Gochujang marinade on inhibition of
86 16 PAHs in charcoal-grilled pork belly cooked to different levels of doneness.

87

88 **Materials and Methods**

89 **Materials**

90 Gochujang (Sunchang Gochujang; Chungjungone Co., Seoul, Korea) used in this study
91 was composed of brown rice (20.4%), red pepper powder (3%), and red pepper seasoning
92 (red pepper powder [8.3%], sea salt, garlic, and onion) with soybeans, alcohol, yeast powder,
93 starch syrup, brown glutinous rice flour, sea salt, and isomaltooligosaccharide. Folin-
94 Ciocalteu phenol reagent was purchased from Sigma Co. (St. Louis, MO, USA). All solvents
95 and chemicals used for PAH analysis were high-performance liquid chromatography grade.
96 PAH standards (naphthalene, acenaphthylene, acenaphthene, fluorene, phenanthrene,
97 anthracene, fluoranthene, pyrene, benzo[a]anthracene, chrysene, benzo[b]fluoranthene,
98 benzo[k]fluoranthene, benzo[a]pyrene, indeno[1,2,3-cd]pyrene, dibenzo[a,h]anthracene, and
99 benzo[g,h,i]perylene) and internal standard (ISTD) solution mix (naphthalene-d₈,
100 acenaphthene-d₁₀, phenanthrene-d₁₀, chrysene-d₁₂, and perylene-d₁₂) were purchased from
101 Sigma Co. All the other chemicals were of analytical grade.

102

103 **Phenol contents in Gochujang**

104 The phenol content was measured using Folin-Ciocalteu colorimetric assays, as described
105 by Singleton et al. (1999). The Gochujang was diluted in methanol, and the diluted sample
106 (0.5 mL) was mixed with distilled water (5 mL) and Folin-Ciocalteu phenol reagent (0.5 mL).
107 The mixed solution was incubated for 3 min at 25°C. Subsequently, 1 mL of 1 N Na₂CO₃ was
108 added, and the solution was incubated for 90 min at 25°C in the dark. The absorbance of the
109 reactant was measured at 760 nm using a spectrophotometer (Molecular Devices, CA, USA).
110 The standard curve was established using gallic acid, and the results were expressed as mg
111 gallic acid equivalent (GAE)/g.

112

113 **Antioxidant activities in Gochujang**

114 Oxygen radical absorption capacity (ORAC)

115 ORAC assays were performed according to the method of Kim et al. (2019b). First, 25 µL
116 diluted sample was mixed with 150 µL of 80 nM fluorescein and incubated for 15 min at
117 37° C. After incubation, to generate peroxy radicals, 25 µL of 150 mM 2,2'-azobis (2-
118 amidinopropane) hydrochloride was added. The change in the absorbance of the reactant was
119 recorded every minute at excitation (480 nm) and emission (520 nm) wavelengths, at 37°C
120 for 60 min using a spectrophotometer (Molecular Devices). Trolox (Sigma Co.) was used to
121 generate a standard curve, and the results were expressed as mmol trolox equivalent (TE)/g.

122

123 2,2-Azinobis (3-ethyl-benzothiazoline-6-sulfonic acid) (ABTS) radical scavenging
124 activity

125 The ABTS radical-scavenging activity was evaluated as described by Kim et al. (2019b).
126 To generate the ABTS⁺ radical, a 14 mM ABTS⁺ and a 4.9 mM potassium persulfate
127 solution were mixed (1:1, v/v), and the resulting solution was then reacted at $23 \pm 1^\circ\text{C}$ for 12
128 h in the dark. The stock solution was diluted to reach an absorbance of 0.700 ± 0.002 at 735
129 nm and 30°C . Next, 50 μL sample was reacted with the ABTS⁺ radical solution (950 μL) for
130 30 min at 30°C in the dark. The absorbance of the solution was determined at 735 nm. The
131 results were expressed as mmol TE/g.

132

133 Ferric reducing antioxidant power (FRAP) activity

134 FRAP assays were carried out according to the method of Kim et al. (2019b). The FRAP
135 working solution was prepared by mixing 300 mM acetate buffer (pH 3.6), 10 mM 2,4,6-
136 tripyridyl-S-triazine in 40 mM HCl, and 20 mM $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ solution at a ratio of 10:1:1
137 (v/v/v), respectively. Each sample (25 μL) was reacted with the FRAP working solution (175
138 μL) for 30 min at 37°C in the dark. The absorbance of the reaction solution was determined at
139 590 nm. The results were expressed as mmol TE/g.

140

141 2, 2-Diphenyl-1-picrylhydrazyl (DPPH) radical scavenging activity

142 The DPPH radical scavenging activity was determined following the method of Kim and
143 Jang (2021). Each sample (100 μL) was reacted with 0.2 mM DPPH solution (100 μL) in a
144 96-well microplate. The mixtures were incubated at 25°C for 30 min in the dark, and
145 absorbance was measured at 517 nm. The results were expressed as mmol TE/g.

146

147 **Preparation of pork belly without marinade (PBW) and pork belly marinated**

148 **with Gochujang (PBG)**

149 The preparation of PBW and PBG is detailed in Fig. 1. Frozen pork belly (LYD) and
150 Gochujang were purchased from a local supermarket in Korea. The pork belly was cut into
151 cubes measuring 10 cm (length) × 5 cm (width) × 0.4 cm (thickness). The pork belly was
152 marinated under the following conditions: 71.94% pork belly, 21.58% Gochujang, and 6.48%
153 water, according to previous sensory evaluation and PAHs analysis (data not shown). PBW
154 and PBG were placed into a polyester bag to ripen under vacuum conditions for 24 h at 5°C.
155 The ripened PBW and PBG were stored under vacuum conditions at $9 \pm 2^\circ\text{C}$ for 10 days and
156 then used for analysis on days 0, 5, and 10.

157

158 **Charcoal grilling of pork belly**

159 Approximately 1 kg charcoal was placed into a garden-type grill (55 cm width, 34 cm
160 length, and 14 cm height; Allcook, Korea). After ignition, grilling was performed over the
161 charcoal at 600°C and at a distance of 8 cm from the heat source. According to internal
162 temperature, doneness of grilled pork belly divided into two; the total grilling times were 3.5
163 min for moderate cooking (internal temperature: 71°C; MC) and 5 min for well-done cooking
164 (internal temperature: 81°C; WC). The internal temperature used in this study fitted on safe
165 temperature for cooking pork meat by USDA guideline (Jang et al., 2019). No oil was applied
166 to the meat surface, and the meat was turned once during cooking.

167

168 **Analysis of PAHs**

169 PAH analysis was carried out as described by Kim et al. (2021). Pork belly samples (2.5 g)
170 were weighed in conical tubes, and 5 mL ethyl acetate/acetonitrile (20:80, v/v) with ISTD

171 mix (naphthalene-d₈, acenaphthene-d₁₀, phenanthrene-d₁₀, chrysene-d₁₂, and perylene-d₁₂, 400
172 ng/mL) was added. The samples were ultrasonicated for 20 min, and after centrifugation
173 (3,000 × g, 7 min), the supernatants were transferred to new conical tubes. Next, 5 mL ethyl
174 acetate/acetonitrile (20:80, v/v) was added to the remaining pellets, and samples were
175 ultrasonicated for 20 min. The combined supernatants were concentrated to 2 mL under
176 vacuum conditions using a rotary evaporator (Scilab Korea, Co., Ltd., Seoul, Korea), and 0.5
177 mL distilled water was added. The extracts (2 mL concentrate + 0.5 mL distilled water, total
178 = 2.5 mL) were cleaned up by passing through a Captiva EMR-Lipid cartridge (Agilent
179 Technologies, Santa Clara, CA, USA). Subsequently, 0.625 mL ethyl
180 acetate/acetonitrile/water (16:64:20, v/v/v) was eluted through the Captiva EMR-Lipid
181 cartridge. After elution, 1.875 mL eluent was transferred to a new conical tube, mixed with
182 2.625 mL distilled water and 1.2 mL isooctane, and shaken vigorously. After centrifugation
183 for 7 min at 3,000 × g, PAH contents in the supernatant were analyzed by gas
184 chromatography/mass spectrometry (Agilent 8890 GC with an Agilent 5977B GC/MSD;
185 Agilent Technologies, USA).

186 The 16 PAHs were separated using a DB-EUPAH capillary column (20 m × 0.18 mm id,
187 0.14 μm thickness; Agilent Technologies). Pure helium (99.999%) was used as the carrier gas
188 at a constant flow rate of 1.2 mL/min. The samples (1 μL) were injected in the splitless mode
189 with an injector temperature of 300°C. The temperature of the mass selective detector was
190 310°C, and the source temperature was 290°C. The oven temperature was initially held at
191 70°C for 1 min, ramped to 190°C at a rate of 30°C/min, ramped to 290°C at a rate of
192 10°C/min and held for 5 min, and then finally ramped to 320°C at a rate of 30°C/min and
193 held for 1 min. The mass spectrometer was operated in electron ionization mode (70 eV), and
194 quantitative data acquisition was performed in selective ion monitoring mode. Representative

195 chromatograms of the 16 PAH standards are shown in Fig. 2. All PAHs were quantified using
196 the relative response factors related to ISTD by nine-point calibration curves (9–2,400
197 ng/mL). The squared correlation coefficients of determination (R^2) of the calibration curves
198 were found to be over 0.99. The limit of detection of 16 PAHs was the range of 0.01–0.09
199 $\mu\text{g}/\text{kg}$. The limit of quantification of 16 PAHs was the range of 0.03–0.28 $\mu\text{g}/\text{kg}$. The average
200 relative recovery of the 16 PAHs was 80.9–119.5% for pork belly. Moreover, the relative
201 standard deviation was 0.57–4.62%.

202

203 **Statistical analysis**

204 All analyses were expressed as means and standard errors of the means (SEMs). Statistical
205 analysis was performed with SAS software v.9.4 (SAS Institute Inc., Cary, NC, USA) using
206 one-way analysis of variance (ANOVA). The interaction between doneness and marinating
207 with regard to PAH formation in charcoal-grilled pork belly was evaluated using two-way
208 ANOVA by SAS program. Significant differences in means were determined using Tukey's
209 tests and results with p values less than 0.05 were considered significant.

210

211 **Results and Discussion**

212 **Phenol contents and antioxidant activities of Gochujang**

213 The phenol contents and antioxidant activity of the Gochujang used in this study are shown
214 in Fig. 3. Gochujang had a phenol content of 1.05 mg GAE/g. Red pepper, one of the main
215 components of Gochujang, contains 1.73 mg GAE/g phenols (Marinova et al., 2005). Thus,
216 Gochujang had phenol contents similar to those of broccoli (1.01 mg GAE/g) and apples

217 (1.04 mg GAE/g) (Marinova et al., 2005), but lower phenol contents than green tea (86.3 mg
218 GAE/g dry matter) (Khokhar et al., 2002) and strawberries (2.44 mg GAE/g) (Marinova et al.,
219 2005).

220 The ORAC activity of Gochujang was 16.62 mmol TE/g, and the ABTS and DPPH radical
221 scavenging activities were 4.30 and 1.77 mmol TE/g, respectively. Additionally, the FRAP
222 activity of Gochujang was 3.24 mmol TE/g. During Gochujang production, various
223 metabolites, such as polyphenols and amino acid, are produced by microbial fermentation;
224 these components have high nutritional value and are important contributors to antioxidant
225 activity in Gochujang (Lee et al., 2016; Yang et al., 2018).

226

227 **PAH contents in charcoal-grilled pork belly marinated with Gochujang**

228 The contents of 16 PAHs in charcoal-grilled PBW and PBG on days 0, 5, and 10 are
229 shown in Tables 1–3. Several factors affect PAH formation in charcoal-grilled meats,
230 including the amount of fat in the meat, the closeness to the heat source, and the cooking time
231 (Chung et al., 2011). In particular, pyrolysis of proteins, fats, and carbohydrates is accelerated
232 by cooking at high temperature, leading to the generation of PAHs (Kılıç Büyükkurt et al.,
233 2020). In this study, WC increased the PAH contents in PBW and PBG compared with MC
234 ($p < 0.0001$). These results were similar to those of Wang et al. (2019) and Kim et al. (2021),
235 who showed that increased doneness accelerated PAH formation in meats. Phenanthrene was
236 the most abundant compound, with contents ranging from 273.60 to 503.02 $\mu\text{g}/\text{kg}$. Among
237 the four major PAHs detected in this study, benzo[a]pyrene was the most abundant
238 compound, with contents ranging from 16.24 to 63.39 $\mu\text{g}/\text{kg}$. In another study,
239 benzo[a]pyrene was detected in charcoal-grilled pork belly at 8.04 $\mu\text{g}/\text{kg}$ (Park et al., 2017),

240 which was lower than the amount reported in this study. Notably, the amount of PAH in
241 charcoal-grilled meat increases when the meat is cooked closer to the charcoal; Park et al.
242 (2017) cooked their pork belly at 15 cm from the charcoal, whereas we cooked our pork belly
243 samples at 8 cm from the charcoal. The total PAH contents in PBW-MC and PBW-WC were
244 1,507.22 and 2,009.45 $\mu\text{g}/\text{kg}$, respectively. However, PBG reduced the total PAH formation,
245 showing contents of 982.45 and 1,918.90 $\mu\text{g}/\text{kg}$ in PBG-MC and PBG-WC, respectively,
246 compared with PBW ($p < 0.05$). Additionally, the contents of the four most abundant PAHs
247 were reduced by PBG (39.86 and 121.26 $\mu\text{g}/\text{kg}$ in PBG-MC and PBG-WC, respectively),
248 compared with PBW ($p < 0.05$). Moreover, the contents of 14 PAHs (excluding
249 benzo[b]fluoranthene, benzo[k]fluoranthene, and indono[1,2,3-cd]pyrene) in charcoal-grilled
250 pork belly were affected by treatment ($p < 0.001$, $p < 0.0001$) and doneness ($p < 0.0001$), and
251 treatment and doneness interacted with each other to affect PAH contents ($p < 0.05$, $p <$
252 0.0001).

253 On day 5, similar to the results on day 0, 14 PAHs were detected in charcoal-grilled PBW
254 and PBG; chrysene and dibenzo[a,h]anthracene were not detected (Table 2). According to the
255 doneness of charcoal-grilling, WC increased the PAH contents (except the content of
256 naphthalene) in PBW and PBG compared with MC ($p < 0.0001$). Naphthalene contents in
257 charcoal-grilled pork belly were not significantly affected by doneness. Phenanthrene was the
258 most abundant compound, with contents ranging from 319.93 to 751.04 $\mu\text{g}/\text{kg}$. Among the
259 four major PAHs, benzo[a]pyrene was the most abundant, showing contents ranging from
260 30.62 to 114.84 $\mu\text{g}/\text{kg}$. The total PAH contents for all 16 PAHs in PBW-MC and PBW-WC
261 were 1,919.37 and 3,013.16 $\mu\text{g}/\text{kg}$, respectively; this was increased compared with that from
262 day 0. However, PBG inhibited the formation of total 16PAHs, showing reduced contents of
263 1189.63 and 1,402.64 $\mu\text{g}/\text{kg}$ for PBG-MC and PBG-WC, respectively, compared with PBW

264 ($p < 0.05$). The contents of the four most abundant PAHs were also reduced in PBG to 70.24
265 and 100.56 $\mu\text{g}/\text{kg}$ for PBG-MC and PBG-WC, respectively, compared with PBW ($p < 0.05$).
266 The contents of 14 PAHs (except naphthalene) in charcoal-grilled pork belly were affected by
267 treatment ($p < 0.0001$) and doneness ($p < 0.05$, $p < 0.0001$). An interaction effect was also
268 observed between treatment and doneness for PAH contents ($p < 0.01$, $p < 0.0001$).

269 On day 10, similar to the results from days 0 and 5, 14 PAHs were detected in charcoal-
270 grilled PBW and PBG; chrysene and dibenzo[a,h]anthracene were not detected (Table 3).
271 WC still increased the contents of all PAHs in PBW and PBG compared with MC ($p < 0.01$,
272 $p < 0.0001$). Phenanthrene was the most abundant compound, with contents ranging from
273 227.88 to 782.97 $\mu\text{g}/\text{kg}$. Among the four major PAHs, benzo[a]pyrene was the most abundant
274 compound, with contents ranging from 20.49 to 134.75 $\mu\text{g}/\text{kg}$. The contents of the total
275 16PAHs in PBW-MC and PBW-WC were 2,254.98 and 3,225.90 $\mu\text{g}/\text{kg}$, respectively,
276 indicating increased contents compared with days 0 and 5. However, PBG inhibited the
277 formation of the total 16PAHs to 883.07 and 11,540.15 $\mu\text{g}/\text{kg}$ for PBG-MC and PBG-WC,
278 respectively, compared with PBW ($p < 0.05$). Additionally, the contents of the four most
279 abundant PAHs were reduced in PBG, showing contents of 49.76 and 107.89 $\mu\text{g}/\text{kg}$ for PBG-
280 MC and PBG-WC, respectively, compared with PBW ($p < 0.05$). For 14 types of PAHs, the
281 contents in charcoal-grilled pork belly were affected by treatment ($p < 0.0001$) and doneness
282 ($p < 0.01$, $p < 0.0001$). The interaction effect between treatment and doneness was observed
283 for some of the PAHs and all four of the major PAHs ($p < 0.001$, $p < 0.0001$), but not for
284 fluorene, phenanthrene, and the total 16PAHs.

285 Few studies have evaluated the effects of storage on PAH formation in grilled meats. In
286 this study, we found that PAH contents in pork belly were increased as the storage time
287 increased except naphthalene and fluorene, which these contents showed an increase on day 5

288 and decrease on day 10. Similar to our results, Zhao et al. (2018) reported that the
289 concentration of PAHs in oil increased with storage time and that increases in oxidation and
290 radical formation during storage could be attributed to increased PAH concentrations.
291 Although the reason for the increase in PAH contents during storage and after charcoal-
292 grilling is not fully understood, our findings suggested that increased oxidation in pork belly
293 during storage could contribute to production of high PAH contents in pork belly on days 5
294 and 10 when charcoal-grilling.

295

296 **Inhibition of PAH content by Gochujang marinade in charcoal-grilled pork** 297 **belly**

298 The Gochujang marinade showed inhibitory effects on PAH formation in charcoal-grilled
299 pork belly during storage (Tables 1–3). The inhibition rates of the three major PAHs
300 (benzo[a]anthracene, B[a]A; benzo[b]fluoranthene, B[b]F; benzo[a]pyrene, B[a]P) and the 16
301 total PAHs were calculated in PBG and PBW (Fig. 4). On day 0, the inhibition rates of MC
302 on each of the three major PAHs and on the total 16PAHs were higher than those of WC
303 (34.82–50.47% versus 4.51–24.30%; $p < 0.05$). The highest and lowest PAH inhibitory
304 effects were observed for B[b]F and the total 16PAHs, respectively. This may be because of
305 the high PAH contents in charcoal-grilled pork belly subjected to WC, results in apparent
306 weakening of the inhibitory effects of Gochujang. However, the inhibition rates of WC on the
307 three major PAHs and the total 16PAHs on day 5 were higher than those of MC ($p < 0.05$).
308 Finally, on day 10, there were no significant differences in inhibition rates between MC and
309 WC for the three major PAHs. Additionally, the inhibition rates of the three major PAHs and
310 the 16 total PAHs increased up to 52.26–67.80% on day 10. These results indicated that the

311 inhibitory activity of Gochujang on PAH contents in charcoal-grilled pork belly was
312 increased as the storage time increased. Similar to our results, García-Lomillo et al. (2017)
313 reported that red wine pomace seasoning on beef patties for 9 days increased the PAH
314 inhibitory effect compared with that on day 1. This suggests that the inhibitory effects of
315 polyphenols on PAH formation are exerted during storage (García-Lomillo et al., 2017).
316 Therefore, storage of pork belly marinated Gochujang for 10 days could increase the PAH
317 inhibitory effect when charcoal-grilling.

318 Previous studies have reported that marinades, such as onions, garlic, beer, tea, and vinegar,
319 can reduce the PAH contents of grilled meats (Cordeiro et al., 2020; Viegas et al., 2014;
320 Wang et al., 2019). These ingredients contain abundant amounts of phenolic compounds and
321 sulfhydryl compounds, resulting in high antioxidant activity. In particular, antioxidants can
322 inhibit PAH formation in meats by blocking free radical generation in pyrolysis and
323 eliminating the free radicals (Cordeiro et al., 2020; Viegas et al., 2014; Wang et al., 2019).
324 Gochujang is fermented by *Aspergillus oryzae*, and during fermentation, metabolites such as
325 furan, phenolics, and heptelidic acid derivatives can be produced, exerting various biological
326 effects (Lee et al., 2016). Moreover, during the fermentation, total polyphenols and
327 flavonoids in Gochujang increase, thereby enhancing its antioxidant activity (Yang et al.,
328 2018). Therefore, the high phenolic content and antioxidant activity of Gochujang can inhibit
329 PAH formation in charcoal-grilled pork belly.

330 The PAH formation during grilling is the result of pyrolysis of nutrients on the surface of
331 meats caused by exposure to high-temperature flame, and incomplete combustion of coal
332 results in adherence of smoke particles to surface of the meat. Therefore, blocking the surface
333 using a marinade can reduce the PAH contents in grilled meats (Kılıç Büyükkurt et al., 2020).
334 Because Gochujang is a paste with high viscosity, it can easily cover the surface of pork belly,

335 which may help block the surface of pork belly from the flame and smoke.

336

337 **Conclusion**

338 The phenol contents and antioxidant activities of Gochujang may affect the reduction of
339 PAH contents in charcoal-grilled pork belly. Cooking at a high internal temperature of 81 °C
340 (well-done cooking) increased the formation of 16 PAHs in charcoal-grilled pork belly
341 compared with moderate cooking at 71 °C. Interactions were observed between doneness and
342 marinating with regard to PAH contents in charcoal-grilled pork belly. Moreover, the
343 inhibitory effects of Gochujang marinade on the total 16PAHs in charcoal-grilled pork belly
344 increased up to 63.06% after storage for 10 days and cooking at moderate temperature.
345 Importantly, the amount of carcinogenic benzo[a]pyrene in charcoal-grilled pork belly
346 marinated with Gochujang was 67.80% on day 10 after well-done cooking. These results
347 indicated that Gochujang marinade of pork belly could inhibit PAH formation after charcoal-
348 grilling and that the inhibitory effects of this marinade increased during the 10 days of
349 storage. These findings provided a preliminary data for the inhibitory effects of Gochujang on
350 PAH formation, while the specific mechanism for PAH inhibitory effect of Gochujang
351 marinade on charcoal-grilled pork belly should be investigated in the future. It was suggested
352 that marinating pork belly with Gochujang may be an effective processing method to reduce
353 the intake of PAHs when consuming charcoal-grilling meat.

354

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361

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433 different red pepper (*Capsicum annuum* L.) varieties. *J Food Sci Technol* 55:792-801.

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435 oxygenated polycyclic aromatic hydrocarbons in crude and refined vegetable oils. *Food*
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ACCEPTED

437 **Table 1. Effect of Gochujang marinade, doneness, and interaction on inhibition of**
 438 **16 PAHs in charcoal-grilled pork belly on day 0**

PAHs ($\mu\text{g}/\text{kg}$)	Doneness	Treatment		SEM	Significance		
		PBW	PBG		T	D	T×D
Naphthalene	MC	141.89 ^{Aa}	65.94 ^{Bb}	2.724			
	WC	142.55 ^{Aa}	111.10 ^{Ab}	2.024	*****	*****	*****
	SEM	2.992	1.601				
Acenaphthylene	MC	339.06 ^{Ba}	290.62 ^{Bb}	4.196			
	WC	375.81 ^{Ab}	459.74 ^{Aa}	4.649	*****	*****	*****
	SEM	3.891	4.907				
Acenaphthene	MC	11.62 ^{Aa}	8.71 ^{Bb}	0.150			
	WC	10.67 ^{Ba}	12.92 ^{Aa}	0.173	*****	*****	*****
	SEM	0.116	0.197				
Fluorene	MC	76.72 ^{Ba}	62.20 ^{Bb}	1.323			
	WC	82.30 ^{Aa}	97.79 ^{Ab}	0.949	**	*****	*****
	SEM	1.174	1.129				
Phenanthrene	MC	417.21 ^{Ba}	273.60 ^{Bb}	5.104			
	WC	503.02 ^{Aa}	473.19 ^{Ab}	4.779	*****	*****	*****
	SEM	5.336	4.519				
Anthracene	MC	52.63 ^{Ba}	34.72 ^{Bb}	0.478			
	WC	63.83 ^{Aa}	73.61 ^{Ab}	0.946	***	*****	*****
	SEM	0.493	0.938				
Fluoranthene	MC	162.52 ^{Ba}	86.35 ^{Bb}	1.260			
	WC	257.26 ^{Aa}	217.14 ^{Ab}	2.002	*****	*****	*****
	SEM	1.199	2.039				
Pyrene	MC	170.39 ^{Ba}	91.34 ^{Bb}	2.258			
	WC	304.42 ^{Aa}	260.20 ^{Ab}	2.607	*****	*****	*****
	SEM	2.149	2.697				
Benzo[a]anthracene	MC	27.82 ^{Ba}	14.28 ^{Bb}	0.443			
	WC	50.79 ^{Aa}	40.57 ^{Ab}	0.785	*****	*****	*
	SEM	0.543	0.720				
Chrysene	MC	ND	ND	-			
	WC	ND	ND	-	-	-	-
Benzo[b]fluoranthene	MC	18.85 ^{Ba}	9.34 ^{Bb}	0.399			
	WC	35.89 ^{Aa}	27.17 ^{Ab}	0.860	*****	*****	ns

	SEM	0.690	0.650				
Benzo[k]fluoranthene	MC	11.66 ^{Ba}	6.02 ^{Bb}	0.260			
	WC	23.32 ^{Aa}	18.14 ^{Ab}	0.407	****	****	ns
	SEM	0.356	0.326				
Benzo[a]pyrene	MC	31.73 ^{Ba}	16.24 ^{Bb}	0.768			
	WC	63.19 ^{Aa}	53.51 ^{Ab}	1.014	****	****	*
	SEM	0.777	1.007				
Indeno[1,2,3-cd]pyrene	MC	19.87 ^{Ba}	10.24 ^{Bb}	0.563			
	WC	42.13 ^{Aa}	31.21 ^{Ab}	0.612	****	****	ns
	SEM	0.594	0.582				
Dibenzo[a,h]anthracene	MC	ND	ND	-	-	-	-
	WC	ND	ND	-	-	-	-
Benzo[ghi]perylene	MC	25.25 ^{Ba}	12.86 ^{Bb}	0.700			
	WC	54.26 ^{Aa}	42.61 ^{Ab}	0.956	****	****	ns
	SEM	0.631	1.003				
Total 16PAHs	MC	1507.22 ^{Ba}	982.45 ^{Bb}	18.931			
	WC	2009.45 ^{Aa}	1918.90 ^{Ab}	18.208	****	****	****
	SEM	19.608	17.477				
4PAHs	MC	78.40 ^{Ba}	39.86 ^{Bb}	1.592			
	WC	149.87 ^{Aa}	121.26 ^{Ab}	2.589	****	****	*
	SEM	1.958	2.325				

439 PBW, pork belly without marinade; PBG, pork belly marinated with Gochujang; T,
440 Treatment; D, Doneness; T×D, Treatment × Doneness; MC, moderate cooking; WC,
441 well-done cooking; ND, not detected. ns, not significantly; *, p<0.05; **, p<0.01;
442 ***p<0.001; ****p<0.0001. 4PAHs : Benzo[a]anthracene + Chrysene +
443 Benzo[b]fluoranthene + Benzo[a]pyrene

444 ^{A-B} Means within a column with different superscript differ significantly at p<0.05.

445 ^{a-b} Means within a row with different superscript differ significantly at p<0.05.

446

447 **Table 2. Effect of Gochujang marinade, doneness, and interaction on inhibition of**
 448 **16 PAHs in charcoal-grilled pork belly on day 5**

PAHs ($\mu\text{g}/\text{kg}$)	Doneness	Treatment		SEM	Significance		
		PBW	PBG		T	D	T×D
Naphthalene	MC	155.17 ^{Aa}	47.02 ^{Ab}	1.293			
	WC	145.40 ^{Aa}	49.87 ^{Ab}	1.656	*****	ns	ns
	SEM	10.397	0.771				
Acenaphthylene	MC	407.10 ^{Ba}	287.93 ^{Ab}	7.074			
	WC	453.24 ^{Aa}	278.56 ^{Ab}	6.032	*****	*	**
	SEM	7.951	4.817				
Acenaphthene	MC	17.49 ^{Aa}	11.53 ^{Ab}	0.248			
	WC	14.61 ^{Ba}	10.87 ^{Ab}	0.356	*****	***	**
	SEM	0.288	0.324				
Fluorene	MC	117.11 ^{Ba}	83.17 ^{Ab}	2.090			
	WC	138.63 ^{Aa}	85.69 ^{Ab}	2.650	*****	***	**
	SEM	3.000	1.546				
Phenanthrene	MC	498.10 ^{Ba}	319.93 ^{Bb}	6.542			
	WC	751.04 ^{Aa}	383.87 ^{Ab}	10.997	*****	*****	*****
	SEM	11.609	5.381				
Anthracene	MC	80.99 ^{Ba}	55.00 ^{Bb}	1.302			
	WC	118.90 ^{Aa}	67.96 ^{Ab}	1.514	*****	*****	*****
	SEM	1.589	1.209				
Fluoranthene	MC	205.23 ^{Ba}	119.91 ^{Bb}	2.211			
	WC	405.03 ^{Aa}	164.56 ^{Ab}	6.211	*****	*****	*****
	SEM	5.344	3.861				
Pyrene	MC	234.44 ^{Ba}	146.87 ^{Bb}	2.707			
	WC	507.13 ^{Aa}	191.03 ^{Ab}	5.063	*****	*****	*****
	SEM	4.970	2.874				
Benzo[a]anthracene	MC	40.21 ^{Ba}	24.55 ^{Bb}	0.873			
	WC	90.37 ^{Aa}	34.50 ^{Ab}	1.135	*****	*****	*****
	SEM	1.310	0.579				
Chrysene	MC	ND	ND	-	-	-	-
	WC	ND	ND	-	-	-	-
Benzo[b]fluoranthene	MC	26.90 ^{Ba}	15.07 ^{Bb}	0.480			

	WC	62.92 ^{Aa}	23.67 ^{Ab}	0.761	****	****	****
	SEM	0.788	0.435				
Benzo[k]fluoranthene	MC	17.38 ^{Ba}	9.35 ^{Bb}	0.553			
	WC	39.88 ^{Aa}	13.16 ^{Ab}	0.604	****	****	****
	SEM	0.776	0.261				
Benzo[a]pyrene	MC	49.64 ^{Ba}	30.62 ^{Bb}	1.188			
	WC	114.84 ^{Aa}	42.39 ^{Ab}	1.668	****	****	****
	SEM	1.795	0.986				
Indeno[1,2,3-cd]pyrene	MC	30.65 ^{Ba}	16.79 ^{Bb}	0.509			
	WC	72.57 ^{Aa}	24.42 ^{Ab}	0.628	****	****	****
	SEM	0.612	0.528				
Dibenzo[a,h]anthracene	MC	ND	ND	-	-	-	-
	WC	ND	ND	-	-	-	-
Benzo[ghi]perylene	MC	38.98 ^{Ba}	21.87 ^{Bb}	0.680			
	WC	98.60 ^{Aa}	32.09 ^{Ab}	1.312	****	****	****
	SEM	1.259	0.775				
Total 16PAHs	MC	1919.37 ^{Ba}	1189.63 ^{Bb}	30.488			
	WC	3013.16 ^{Aa}	1402.64 ^{Ab}	31.898	****	****	****
	SEM	40.947	16.441				
4PAHs	MC	116.75 ^{Ba}	70.24 ^{Bb}	2.513			
	WC	268.13 ^{Aa}	100.56 ^{Ab}	3.458	****	****	****
	SEM	3.798	1.963				

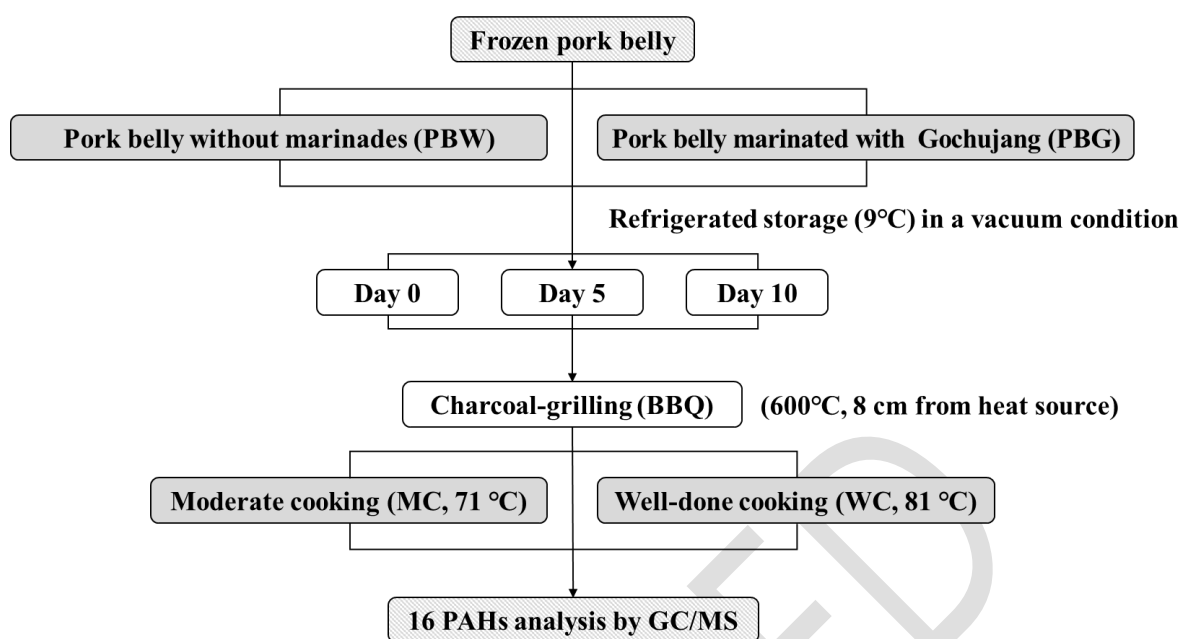
449 PBW, pork belly without marinade; PBG, pork belly marinated with Gochujang; T,
450 Treatment; D, Doneness; T×D, Treatment × Doneness; MC, moderate cooking; WC,
451 well-done cooking; ND, not detected. ns, not significantly; *, p<0.05; **, p<0.01;
452 ***p<0.001; ****p<0.0001. 4PAHs : Benzo[a]anthracene + Chrysene +
453 Benzo[b]fluoranthene + Benzo[a]pyrene
454 ^{A-B} Means within a column with different superscript differ significantly at p<0.05.
455 ^{a-b} Means within a row with different superscript differ significantly at p<0.05.

456 **Table 3. Effect of Gochujang marinade, doneness, and interaction on inhibition of**
 457 **16 PAHs in charcoal-grilled pork belly on day 10**

PAHs ($\mu\text{g}/\text{kg}$)	Doneness	Treatment		SEM	Significance		
		PBW	PBG		T	D	T×D
Naphthalene	MC	189.78 ^{Aa}	61.90 ^{Bb}	4.611			
	WC	130.70 ^{Ba}	87.95 ^{Ab}	5.133	****	**	****
	SEM	5.156	4.585				
Acenaphthylene	MC	536.16 ^{Aa}	191.23 ^{Bb}	11.049			
	WC	523.94 ^{Aa}	285.67 ^{Ab}	9.870	****	**	***
	SEM	9.016	11.757				
Acenaphthene	MC	15.97 ^{Aa}	4.16 ^{Bb}	0.426			
	WC	15.74 ^{Aa}	9.19 ^{Ab}	0.310	****	***	***
	SEM	0.205	0.486				
Fluorene	MC	109.46 ^{Ba}	42.58 ^{Bb}	3.643			
	WC	147.72 ^{Aa}	88.45 ^{Ab}	2.525	****	****	ns
	SEM	3.465	2.764				
Phenanthrene	MC	555.98 ^{Ba}	227.88 ^{Bb}	13.132			
	WC	782.97 ^{Aa}	425.57 ^{Ab}	14.611	****	****	ns
	SEM	13.751	14.031				
Anthracene	MC	82.37 ^{Ba}	21.51 ^{Bb}	1.487			
	WC	138.47 ^{Aa}	53.75 ^{Ab}	2.976	****	****	***
	SEM	2.993	1.454				
Fluoranthene	MC	246.78 ^{Ba}	89.47 ^{Bb}	5.717			
	WC	424.89 ^{Aa}	181.96 ^{Ab}	10.186	****	****	***
	SEM	10.343	5.427				
Pyrene	MC	283.32 ^{Ba}	104.54 ^{Bb}	6.754			
	WC	524.26 ^{Aa}	217.93 ^{Ab}	12.998	****	****	***
	SEM	13.311	6.114				
Benzo[a]anthracene	MC	45.94 ^{Ba}	16.96 ^{Bb}	1.191			
	WC	106.04 ^{Aa}	38.33 ^{Ab}	2.817	****	****	****
	SEM	2.870	1.056				
Chrysene	MC	ND	ND	-			
	WC	ND	ND	-	-	-	-
Benzo[b]fluoranthene	MC	30.65 ^{Ba}	12.31 ^{Bb}	0.918			

	WC	70.61 ^{Aa}	26.17 ^{Ab}	2.154	****	****	****
	SEM	2.198	0.809				
Benzo[k]fluoranthene	MC	20.92 ^{Ba}	8.18 ^{Bb}	0.768			
	WC	46.05 ^{Aa}	17.70 ^{Ab}	1.108	****	****	****
	SEM	1.197	0.621				
Benzo[a]pyrene	MC	58.37 ^{Ba}	20.49 ^{Bd}	1.853			
	WC	134.75 ^{Aa}	43.39 ^{Ad}	3.714	****	****	****
	SEM	3.923	1.356				
Indeno[1,2,3-cd]pyrene	MC	34.88 ^{Ba}	14.68 ^{Bb}	1.133			
	WC	78.25 ^{Aa}	28.35 ^{Ab}	2.733	****	****	***
	SEM	2.794	0.973				
Dibenzo[a,h]anthracene	MC	ND	ND	-	-	-	-
	WC	ND	ND	-	-	-	-
Benzo[ghi]perylene	MC	44.41 ^{Ba}	17.18 ^{Bb}	1.601			
	WC	101.50 ^{Aa}	35.75 ^{Ab}	3.375	****	****	****
	SEM	3.537	1.202				
Total 16PAHs	MC	2254.98 ^{Ba}	833.07 ^{Bb}	48.191			
	WC	3225.90 ^{Aa}	1540.15 ^{Ab}	70.014	****	****	ns
	SEM	67.566	51.566				
4PAHs	MC	134.96 ^{Ba}	49.76 ^{Bb}	3.927			
	WC	311.41 ^{Aa}	107.89 ^{Ab}	8.657	****	****	****
	SEM	8.948	3.211				

458 PBW, pork belly without marinade; PBG, pork belly marinated with Gochujang; T,
459 Treatment; D, Doneness; T×D, Treatment × Doneness; MC, moderate cooking; WC,
460 well-done cooking; ND, not detected. ns, not significantly; *, p<0.05; **, p<0.01;
461 ***p<0.001; ****p<0.0001. 4PAHs : Benzo[a]anthracene + Chrysene +
462 Benzo[b]fluoranthene + Benzo[a]pyrene
463 ^{A-B} Means within a column with different superscript differ significantly at p<0.05.
464 ^{a-b} Means within a row with different superscript differ significantly at p<0.05.

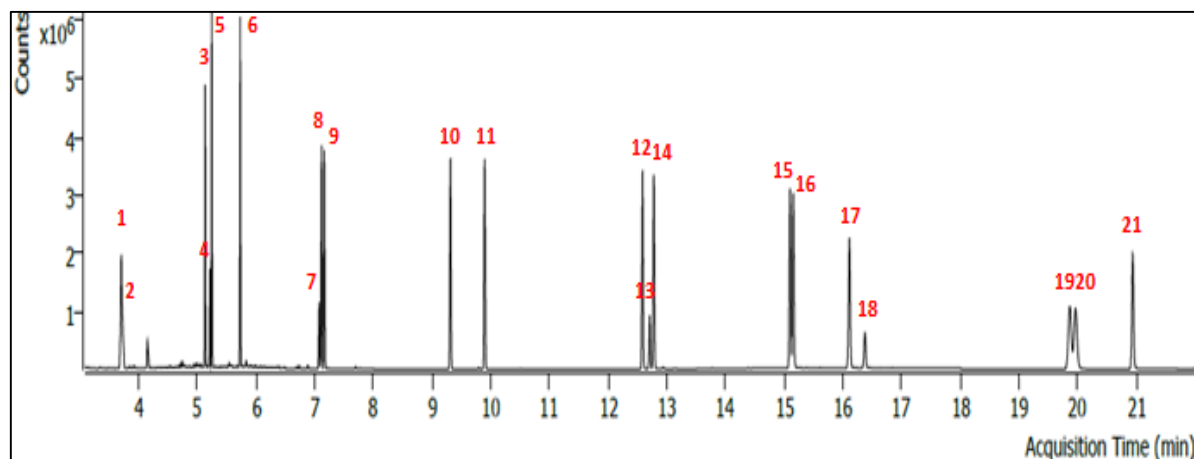


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Fig. 1. Diagram of sample preparation procedure.



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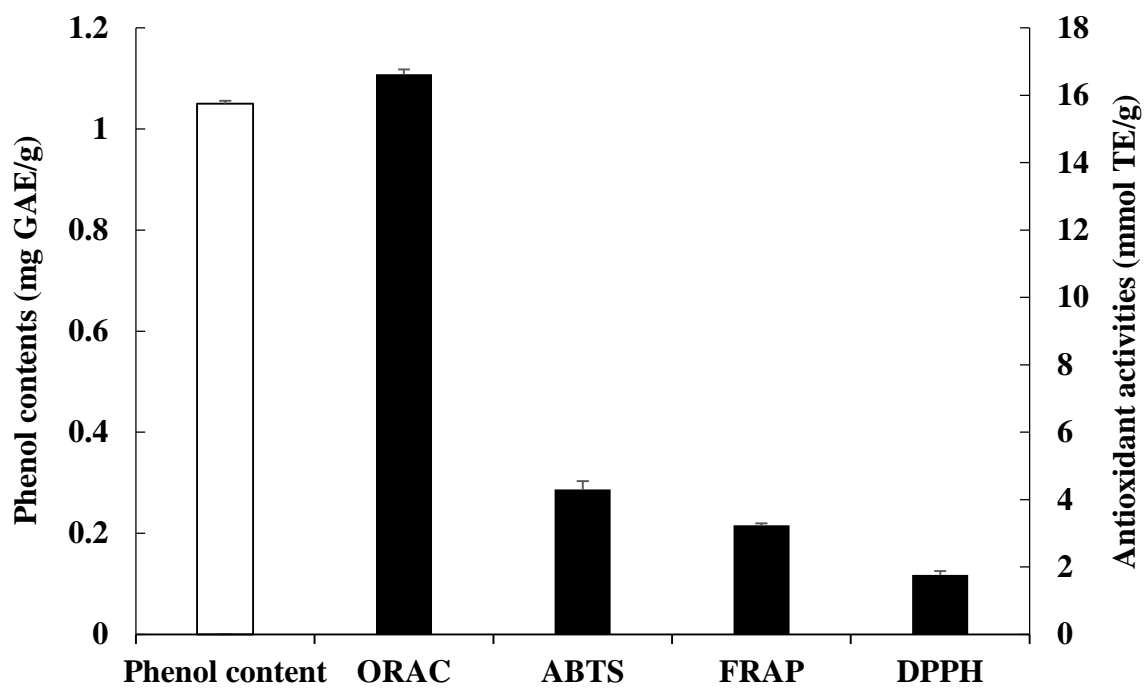
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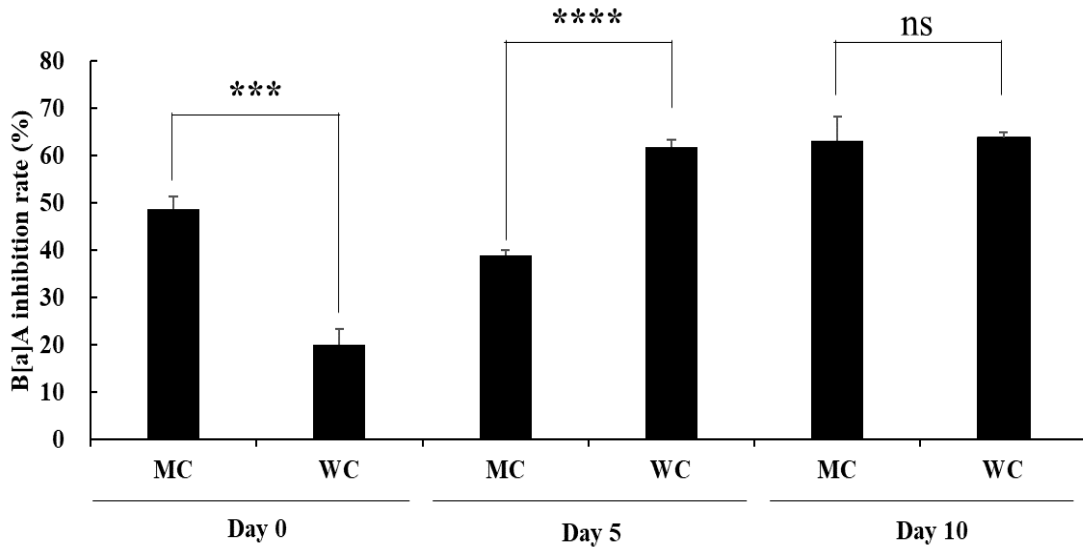
Fig. 2. GC/MS-SIM chromatogram of 16PAHs standards. 1, Naphthalene-d8 (ISTD); 2, Naphthalene; 3, Acenaphthylene; 4, Acenaphthene-d10 (ISTD); 5, Acenaphthene; 6, Fluorene; 7, Phenanthrene-d10 (ISTD); 8, Phenanthrene; 9, Anthracene; 10, Fluoranthene; 11, Pyrene; 12, Benzo[a]anthracene; 13, Chrysene-d12 (ISTD); 15, Benzo[b]fluoranthene; 16, Benzo[k]fluoranthene; 17, Benzo[a]pyrene; 18, Perylene-d12 (ISTD); 19, Indeno[1,2,3-cd]pyrene; 21, Benzo[ghi]perylene.



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Fig. 3. The phenol content and antioxidant activities of Gochujang.

482 (A)

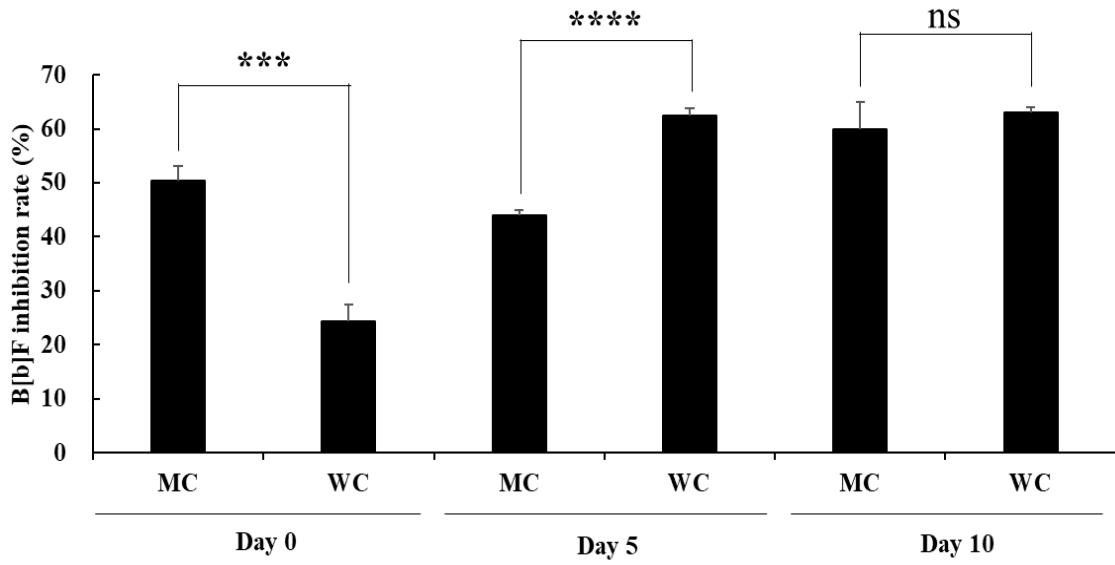


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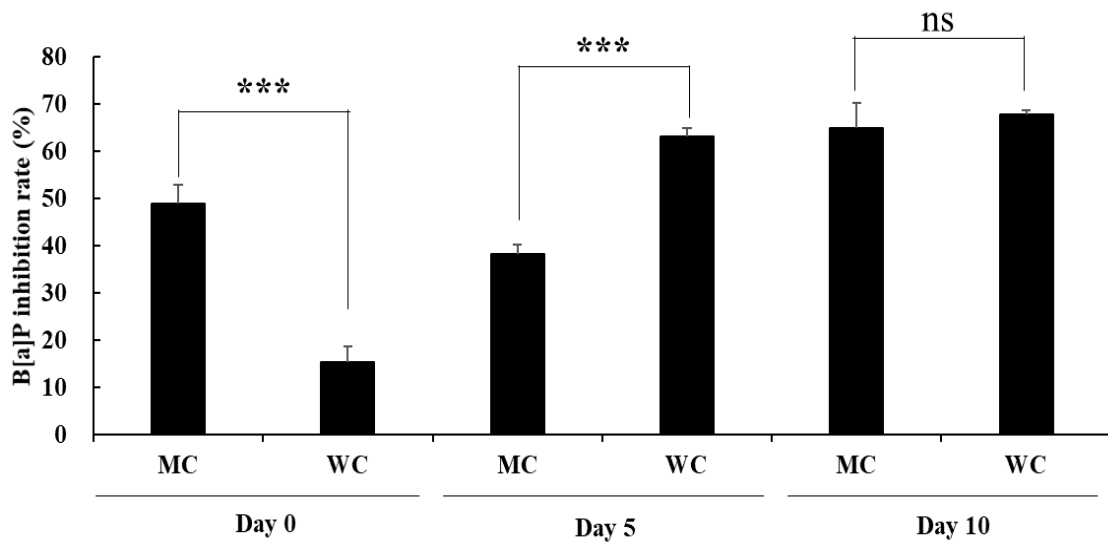
(B)



486

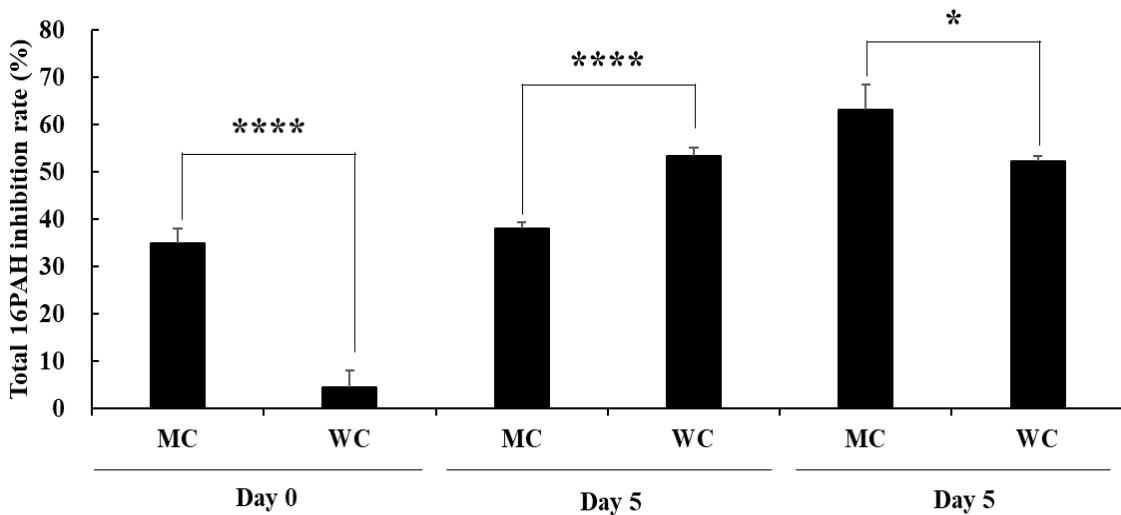
487

488 (C)



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(D)



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Fig. 4. Percentage of inhibition of major three (A-C) and total 16PAHs (D) formation in charcoal-grilled marinated pork belly with Gochujang. B[a]A, benzo[a]anthracene; B[b]F, benzo[b]fluoranthene; B[a]P, benzo[a]pyrene. The inhibition rate was calculated by PAHs contents in pork belly marinated with Gochujang compared to that in pork belly without marinade. MC, moderate cooking; WC, well-done cooking. ns, not significantly; *, $p < 0.05$; * $p < 0.001$; **** $p < 0.0001$.**